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Whey-pectin microcapsules improve the stability of grape marc phenolics during digestion

3 Aimara V. De La Cruz-Molina¹, Catarina Gonçalves², Mafalda D. Neto², Lorenzo

4 Pastrana², Paula Jauregi ^{1,3,4}, Isabel R Amado^{2*}

¹Department of Food and Nutritional Sciences, University of Reading, Whiteknights, Reading, UK,
RG6 6A

7 ²INL- International Iberian Nanotechnology Laboratory, Av. Mestre José Veiga s/n, 4715-330

8 Braga, Portugal

⁹ ³Current address: AZTI, Food Research, Basque Research and Technology Alliance (BRTA),

10 Parque Tecnológico de Bizkaia, Astondo Bidea, Edificio 609, Derio, Bizkaia 48160, Spain

11 ⁴Current address: Ikerbasque, Basque Foundation for Science, Bilbao 48013, Spain

12 * Correspondence: Isabel R. Amado: isabel.rodriguez@inl.int

13 Abstract

14 Grape marc is an agri-food residue from the wine industry valuable for its high content 15 of phenolic compounds. This study aimed to develop an encapsulation system for grape 16 marc extract (GME) using food-grade biopolymers resistant to gastric conditions for its 17 potential use as a nutraceutical. For this purpose, a hydroalcoholic GME was prepared 18 with total phenolics content of 219.62 ± 11.50 mg gallic acid equivalents (GAE)/ g dry 19 extract and 1389.71 \pm 97.33 µmol Trolox equivalents (TE)/ g dry extract antioxidant 20 capacity, assessed through ABTS (2,2'-Azino-bis(3-ethylbenzothiazoline-6-sulfonic 21 acid) assay. Moreover, the extract effectively neutralised reactive oxygen species (ROS) 22 in Caco-2 cells, demonstrating an intracellular antioxidant capacity comparable to Trolox. 23 The GME was encapsulated using whey protein isolate and pectin through nano-spray 24 drying (73% yield), resulting in spherical microparticles with an average size of 1 ± 0.5

25 µm and a polydispersity of 0.717. The encapsulation system protected the microcapsules 26 from simulated gastrointestinal digestion, where at the end of the intestinal phase, 82% 27 of the initial phenolics were bioaccesible compared to 54% in the free GME. Besides, the 28 encapsulated GME displayed a higher antioxidant activity by the ferric reducing 29 antioxidant power (FRAP) assay than the free extract after gastrointestinal digestion 30 (GID). These results show the potential of this encapsulation system for applying GME 31 as a nutraceutical with a high antioxidant capacity and protective effect against cellular 32 oxidation.

- 33 Keywords: grape marc phenolics, biopolymer, nano-spray drying, *in vitro* digestion,
- 34 encapsulation.
- 35

37 **1. Introduction**

38 Grape marc is a food by-product composed of the skins, seeds and stems recovered at the 39 end of the winemaking process. This by-product has attracted significant attention due to

40 its high phenolic content (Lavelli et al., 2016; Peixoto et al., 2018). Phenolics are a family
41 of molecules with antioxidant properties, including phenolic acids and polyphenols such
42 as flavonols and flavan-3-ols (Cao et al., 2021; Tsao, 2010). It has been found that they
43 can play a significant role in the management and prevention of several diseases,
44 especially cardiovascular and type 2 diabetes (Dias et al., 2022; Fraga et al., 2019)

45 Phenolics are extensively researched for their properties but are challenging molecules. 46 They are susceptible to temperature changes, moisture, oxygen, and high/low pH values. 47 In addition, once ingested, they present low stability and bioavailability in the human 48 body due to their low solubility and low membrane permeability (Ludwig et al., 2015; 49 Scalbert & Williamson, 2000; Stalmach et al., 2009; Teng & Chen, 2019). For these 50 reasons, phenolics are unlikely to be used in their pure form and encapsulation is foreseen 51 as an alternative to improve their stability and preserve their properties within food 52 products and bioavailability after consumption (Brezoiu et al., 2019; Sessa et al., 2013; 53 Spigno et al., 2013). The encapsulation process involves using materials to embed, 54 complex, or create a protective wall around bioactives, and by carefully selecting these 55 materials, a targeted release of the bioactives can be achieved.

56 Polysaccharides and proteins are vastly used biopolymers for encapsulation, and 57 interestingly, many of these materials can be obtained from by-products, like whey 58 protein isolate (WPI). WPI is a by-product of the cheese-making process, which contains 59 proteins with high nutritional quality (de Wit, 1998; Jauregi & Welderufael, 2010; Yalçin, 60 2006). Furthermore, WPI forms complexes with polyphenols, stabilising them by

61 improving their solubilisation and protecting their antioxidant activity from heat-induced 62 loss (Guo & Jauregi, 2018). On the other hand, polysaccharides like pectin are found in 63 the peel of citrus, apple, and other fruits. Pectin, as insoluble fibre, is poorly absorbed in 64 the upper gastrointestinal tract (GIT), but pectinolytic enzymes produced by colonic 65 microflora degrade the polysaccharide (Dongowski & Anger, 1996; Rehman et al., 2019). 66 Pectin biodegradability is an interesting property to take advantage of as an effective 67 carrier for the targeted release of bioactive compounds absorbed in the colon. Polyphenols 68 can be absorbed in different parts of the GIT, and those reaching the colon are known to 69 be metabolised by the microbiota into additional low molecular weight phenolic acids 70 (Scalbert et al., 2002). Besides, pectin has other interesting technological properties like 71 emulsifying, gelling and complexation properties (Rehman et al., 2019). In particular, 72 pectin is known for its interaction with WPI through covalent/non-covalent interactions, 73 and their complexes have been studied for their application in food colloidal systems (Du 74 et al., 2022). All these properties of pectin and WPI, together with their known interaction 75 with polyphenols, are expected to protect these labile compounds from processing and 76 digestive conditions, providing their selective release in the lower intestine where they 77 can be absorbed.

78 Among the most used encapsulation methods is spray drying, an efficient, fast, cost-79 effective, and protective method to obtain dry particles (Annunziata et al., 2020; De La 80 Cruz-Molina et al., 2021; Fang & Bhandari, 2012). This encapsulation technique involves 81 the formation of microcapsules by producing a mixture of bioactive compounds with 82 carriers in solution or suspension and then atomising this mixture in a hot air stream to 83 obtain a dry powder (Dias et al., 2022). Nano spray drying (NSD) has emerged as a 84 technology to reduce particle size. With smaller particles, physiological fate is 85 significantly enhanced due to the higher surface: volume ratio offering a higher penetration rate into the cells, stability, target release and bioavailability (Chopde et al.,
2020; Jafari et al., 2021)

88 Several studies have been carried out to study the use of these protein-polysaccharide 89 interactions for spray drying of grape by-products and further in vitro digestion due to the 90 excellent source of phenolics they represent (Brown Da Rocha & Zapata Noreña, 2020; 91 Constantin et al., 2021; Du et al., 2022). However, few studies have investigated nano 92 spray drying for raw extracts and their behaviour during gastrointestinal digestion. Desai 93 et al. (2020) used nano spray drying to encapsulate a raw green coffee extract with 94 maltodextrin; their findings showed that maltodextrin protected the chlorogenic acid and 95 its antioxidant activity from digestion conditions and storage. Other works have used the 96 nano spray dryer for the encapsulation of saffron and soy extracts; however, in these 97 works, a purification of specific compounds was carried out before the encapsulation (Del 98 Gaudio et al., 2016; Kyriakoudi & Tsimidou, 2018). Moreover, these mentioned studies 99 investigate only the use of maltodextrin even though nano spray drying has been used for 100 encapsulation of specific whey proteins such as bovine serum albumin and lactoferrin 101 (Bourbon et al., 2020; Lee et al., 2011).

This work aims to produce nano spray dried microcapsules with whey protein-pectin as encapsulants for the encapsulation of a raw grape marc extract and to study the effect on the stability and bioaccessibility of the polyphenols. Moreover, the biocompatibility and antioxidant capacity of the extract are assessed using a Caco-2 cell line and compared against the commercial antioxidant compound Trolox.

107

108 **2.** Materials and methods

110 Casa Emma Winery (Firenze, Italy) kindly supplied commercial grape marc flour from 111 Sangiovese grapes. The grape marc flour is obtained by drying the grape marc at 42 °C 112 for three days to preserve the phenolics. The grape marc is constantly mixed to avoid 113 mould growth, and after the drying process, it is pulverised to a 250-micron particle size. 114 The final product has the following specifications (supplied by the manufacturer): 8.53% 115 moisture, 8% carbohydrates (from which sugars are 0.56%), 58.6% fibre, and 11.8% 116 protein. Whey protein isolate was purchased from Volac International Ltd (Hertfordshire, 117 UK) with the following specifications (supplied by the manufacturer): protein: 92% min, 118 lactose: 0.9% max, fat: 0.8% max, pH: 5.8 min (10% sol). Pectin from citrus peel with 119 \geq 74.0% of galacturonic acid and \geq 6.7% of methoxy groups; pepsin from porcine gastric 120 mucosa ≥ 250 units/mg solid, pancreatin from porcine pancreas 8 x USP, bile, 2,4,6-121 Tris(2-pyridyl)-s-triazine (TPTZ), 2,2'-Azino-bis(3-ethylbenzothiazoline-6-sulfonic 122 acid) diammonium salt (ABTS), fetal bovine serum (FBS) Superior, Hanks' balanced salt 123 solution (HBSS) Resazurin sodium salt, 2',7'-Dichlorofluorescin diacetate (DCFH-DA), 124 3-Morpholinosydnonimine (Sin-1), (±)-6-Hydroxy-2,5,7,8-tetramethylchromane-2-125 carboxylic acid (Trolox), and tert-butyl hydroperoxide (tBOOH) were purchased from 126 Sigma-Aldrich. Minimum essential medium Eagle (MEM) (with 2 mM L-Glutamine, 1 127 mM Sodium pyruvate, non-essential amino acids (NEAA)) and Penicillin-Streptomycin 128 (10,000 U/mL-10 mg/mL, respectively) were from PAN-Biotech GmbH.

129

130 **2.1 Extraction of phenolics from grape marc**

131

A hydroalcoholic extraction was applied following the methodology previously
developed in our group (MohdMaidin et al., 2018) to extract phenolics from grape marc.
The extraction was carried out in an 8:1 ratio (solvent: solid) using a solution of 60%

ethanol under magnetic stirring for 2 h at 60°C. After the extraction, the solids were
separated through vacuum filtration using No. #1 Whatman paper. Later, the ethanol was
removed from the extract using a rotavapor (RV 10 auto pro-V-C Complete, IKA,
Staufen, Germany). Then, the grape marc extract (GME) was freeze-dried and stored at 18 °C for further analysis, described in sections 2.6 and 2.7.

140

141 **2.2** *In vitro* cell culture studies

142 Cell culture

143 Caco-2 cell line (ATCC, HTB-37) from human colon epithelial carcinoma was routinely 144 expanded in MEM, supplemented with 20% FBS, and 1% Penicillin/Streptomycin (final 145 concentration of 100 U/mL and 100 µg/mL, respectively). The cells were kept in a 146 humidified atmosphere of 5% CO₂, at 37 °C, in 75 cm² flasks. Cells were used in passages 147 33–52, being the cell culture media replaced every other day. Upon reaching confluency, 148 cells were detached using 0.25% trypsin- ethylenediaminetetraacetic acid (EDTA) 149 solution, then pelleted by centrifugation at 300 $\times g$ for 5 min and resuspended in fresh MEM at a concentration of 1×10^5 cells·mL-1. Cells were seeded onto 96-well plates at 150 a density of 1×10^4 cells (100 µL of cellular suspension) per well and left to adhere for 151 152 over 24 h.

153

154 Cell viability assay

The cytotoxicity of GME was determined indirectly by the resazurin conversion assay.After adhesion, the culture medium was removed, cells were washed twice with pre-

157 warmed phosphate buffered saline (PBS) solution, and 200 µL of samples or controls 158 were applied and incubated for 24 h. GME was prepared as described in section 2.1, then 159 further diluted with culture medium (10%, v/v) and tested at 33, 67 and 100 GAE μ g/mL 160 final concentrations based on total phenolic content (TPC) in GME. These concentrations 161 were chosen based on preliminary studies using concentrations reported by Freitas et al. 162 (2020). Negative control was performed using cells growing in MEM (considered 100% 163 cell viability), and 40% (v/v) dimethyl sulfoxide (DMSO) was used as a positive control. 164 After incubation, samples or controls were removed and washed twice with pre-warmed 165 PBS. After this, 100 μ L of 10% (v/v) resazurin in the culture medium (0.01 mg/mL final 166 concentration) was added. The fluorescence intensity, proportional to the number of 167 viable cells, was measured after 5 h of incubation using a microplate fluorescence reader 168 (Synergy H1, BioTek, Vermont, USA) at an excitation wavelength of 560 nm and an 169 emission wavelength of 590 nm. The % cell viability was expressed as the fluorescence 170 of treated cells compared to that of cells growing in the culture medium.

171 Intracellular reactive oxygen species (ROS) quantification

172 The antioxidant activity of GME was determined in an *in vitro* cell assay using DCFH-173 DA as a cell-permeable probe to detect intracellular ROS. After cell adhesion, the culture 174 medium was removed, and 100 µL of 10 µM DCFH-DA solution was added to each well 175 and incubated for 1 h. Afterwards, the solution was removed, and 100 μ L of GME 176 solubilised in HBSS was added to each well at a final concentration of 33 and 67 GAE 177 µg/mL, based on TPC content in GME, and incubated for 4 h. The fluorescence intensity 178 was measured using a microplate fluorescence reader (Synergy H1, BioTek, Vermont, 179 USA) at an excitation wavelength of 495 nm and an emission wavelength of 525 nm.

180 Cells exposed to HBSS, Sin-1 (5 µM) and Trolox (50 µg/mL) were used as basal, positive,
181 and negative controls, respectively.

182 Then, the protective effect of GME against oxidative stress was investigated using Sin-1 183 as an oxidative stress inducer. First, Caco-2 cells were exposed to GME at a 33 and 67 184 GAE µg/mL concentration based on TPC content in GME for 4 h. Then, Sin-1 was added 185 to the cells at a final concentration of 5 µM and incubated for 1 h. The fluorescence 186 intensity was measured every 15 min using a microplate fluorescence reader (Synergy 187 H1, BioteK) at an excitation wavelength of 495 nm and an emission wavelength of 525 188 nm. Cells exposed to HBSS, Sin-1 (5 µM) and Trolox (50 µg/mL) were used as basal, 189 positive, and negative controls, respectively.

190

191 **2.3 Nano-spray drying (NSD)**

192 First, 50 mL of 4% WPI and 0.4% pectin solutions were prepared separately and 193 solubilised overnight at room temperature to ensure complete hydration. Then, 550 mg 194 of GME was resuspended in the pectin solution (50 mL) and mixed with a magnetic stirrer 195 for 5 min. This solution (pectin-GME) was mixed with the WPI solution (50 ml) and 196 stirred for 10 min (magnetic stirring). Then the WPI-pectin-GME solution was 197 centrifuged to remove any large undissolved particles and filtrated through a 0.45 µm 198 PVDF filter before passing it through the NSD. The final solution had a final 199 concentration of 2% WPI, 0.2% pectin and 0.55% GME. A solution containing the same 200 proportion of WPI and pectin, but no GME was prepared to compare physical 201 characteristics. The encapsulation was performed using a Nano-spray Dryer B-90 202 (BÜCHI Labortechnik AG, Flawil, Switzerland). Compressed air was used as the drying 203 gas, and the flow rate was set to about 100 or 110 L/min. The inlet temperature was set

- to 90°C, the spray rate to 65%, and the pump to 30%. WPI-pectin-GME (W-P-GME) and
- 205 WPI-pectin (W-P) particles were stored at 4 °C.
- 206 **2.4 Characterisation of the microparticles**
- 207

208 Scanning electron microscopy (SEM)

209

210 The samples' surface morphology was evaluated through SEM using a Quanta FEG 650

211 (FEI, Oregon, USA). Dried samples were affixed on aluminium stubs covered by carbon

- ribbon and coated with gold, and samples were observed using an accelerating voltage of
- 213 5 kV under vacuum conditions.
- 214 Size and polydispersity index

The size of the particles was determined by analysing SEM images with the program ImageJ (National Institutes of Health, Maryland, USA). The scale was adjusted according to the parameters from SEM images, and the size of 175 particles was determined. After this, the mean and standard deviation was calculated, and from those values, the polydispersity index (PDI) was calculated with the following formula:

220 Equation 1

221
$$PDI = \sqrt{\frac{size \ \sigma}{size \ \bar{x}}}$$

222 Where σ is the standard deviation of the particle size and x is the mean size of the 223 particles.

224

226 Yield

The drying yield was calculated from the ratio of total solids out (microcapsules) to total
solids in (solids in extracts + encapsulants).

229 Equation 2

$$EY\% = \frac{Total \ solids \ out}{Total \ solids \ in} \times \ 100$$

231

232 **Z-potential**

The particles' surface charge (Z- potential) was measured by dynamic light scattering
using an SZ-100 particle analyser (Horiba Scientific, Kyoto, Japan). Microparticles (1
mg/mL) were measured at 25 °C using a He-Ne laser (633 nm) in folded capillary cells.
Five independent measurements of each sample were done, and data were expressed as
mean ±SD.

238 Fourier Transform Infrared Spectroscopy

Fourier Transform Infrared (FTIR) Spectroscopy determined functional groups and the bonding arrangement of sample constituents. FTIR analyses were carried out with an ALPHA II (Bruker, Ettlingen, Germany) spectrometer with a diamond composite in the 400–4000 cm⁻¹ wavenumber region.

243 2.5 In vitro digestion

Particles were tested under simulated digestive conditions to evaluate the protective effect of polymeric particles on GME's activity and polyphenol content. First, the activity of the digestive enzymes (pepsin and trypsin in pancreatin) was quantified. Then, the experimental conditions were applied according to the *in vitro* static INFOGEST method (Brodkorb et al., 2019). The addition of gastric lipase was omitted due to the limited access to the commercially available enzyme, and amylase was not used in the oral phasesince there was no starch in the sample.

251 W-P-GME particles (200 mg) or free GME (100 mg) were resuspended in 1 mL of 252 distilled water and digested. The sample was diluted 1:1 (v/v) in oral digestion with 253 simulated salivary fluid, CaCl₂ 0.3 M and water. The tubes were incubated in an orbital 254 incubator (Fisher Scientific) for 2 min at 37 °C and 150 rpm. For gastric digestion (GD), 255 a pepsin solution (2000 U/mL) in water was prepared based on the previously determined 256 activity. The 2 mL of oral phase were diluted 1:1 (v/v) with simulated gastric fluid, pepsin 257 solution, CaCl₂ 0.3 M, HCl 1 M (to pH 3.0) and water. The samples were incubated for 2 258 h at 37 °C and 150 rpm. A 1.8 mL sample was collected after the 2 h of GD. For intestinal 259 digestion (ID), bile solution and pancreatin were prepared in simulated intestinal fluid. 260 The 2.2 mL of gastric phase were diluted 1:1 (v/v) with simulated intestinal fluid, 261 pancreatin solution, bile, CaCl₂ 0.3 M, NaOH 1 M (to pH 7.0) and water. The samples 262 were incubated for 2 h at 37 °C and 150 rpm. Then the samples were put in an ice water 263 bath for 30 min to stop the enzyme's activity.

After digestion, each digested sample was centrifugated in a Ministar blueline microcentrifuge (fixed speed 2,000 \times *g*) at room temperature for 5 min. The supernatants were collected and stored for analysis. Digestion of polyphenols was evaluated according to the analytical determinations described in sections 2.6 and 2.7 after GD and after gastrointestinal digestion (GID).

The residual values of polyphenols were calculated as a percentage of the total mass of TPC (mg) remaining after GD and after the overall GID in relation to the initial mass. In the case of the antioxidant capacity, the values correspond to the trolox equivalents (TE)

272 (mg) for ABTS and ascorbic acid equivalents (AAE) (mg) for FRAP remaining after each

273 phase of the digestion in relation to the initial ones.

274

275 **2.6 Analytical determinations**

276

277 Total Phenolic Content

278 The total phenolic content (TPC) was determined by the Folin-Ciocalteu method 279 (Singleton & Rossi, 1965). For the assay, 75 µL de Folin-Ciocalteu reagent (1:10) was 280 added in a 96-well microplate, with 15 μ L of the sample and 60 μ L of 7.5% Na₂CO₃. The 281 samples were incubated in the dark for 30 min. After this time, the microplate was read 282 at 765 nm in a microplate reader (Synergy H, BioTek, Vermont, USA). The results were 283 quantified from a Gallic acid calibration curve ranging from 0.1 to 1.0 mg/ml and 284 expressed as milligrams of gallic acid equivalents (GAE) per gram of dried extract (mg 285 GAE / g de).

286

287 Total Monomeric Anthocyanin Content

Total monomeric anthocyanins content (TMAC) levels were quantified by the AOAC Official Method 2005.02 pH differential method (Lee et al., 2005). A sample of GME was combined in a 1:20 ratio (v:v) with potassium chloride and sodium acetate buffers (pH 1.0 and 4.5, respectively) separately. After an equilibration period of 15 min, the absorbance of each solution was measured at 520 and 700 nm in a microplate reader (Synergy H, BioTek, Vermont, USA). The values were calculated with the following formula.

| 295 | Equation 3 |
|------------------------------------------------------------------------------------------|--------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| 296 | Monomeric Anthocyanins = $\frac{A \times MW \times DF \times 1000}{\varepsilon \times 1}$ |
| 297 | Where: |
| 298 | - A= corrected absorbance value calculated as $[(A_{520} - A_{700})_{pH 1.0} - (A_{520} - A_{700})_{pH}]$ |
| 299 | 4.5] |
| 300 | - MW= molecular weight of malvidin 3-O-glucoside (493.43 g/mol) |
| 301 | - DF= dilution factor |
| 302 | - ε = molar absorption: 28,000 L/mol • cm |
| 303 | The results were expressed as milligrams of malvidin 3-O-glucoside equivalents per litre |
| 304 | (mg M3GE/L) |
| 305 | |
| 306 | Total Flavonoid Content |
| 307 | The total flavonoid content (TFC) was measured using the aluminium method (Zhishen |
| 308 | et al., 1999) with some modifications. Briefly, 100 μL of the sample were added to an |
| 309 | Eppendorf tube, and 430 μL of solution A (1.8 mL of 5% NaNO2 mixed with 24 mL of |
| 310 | distilled water) was added to the sample and incubated for 5 min. Later 30 μL of 10% |
| | |
| 311 | AlCl ₃ were added and left to rest for 1 min. Finally, 440 μL of solution B (12 mL of NaOH |
| 311312 | AlCl ₃ were added and left to rest for 1 min. Finally, 440 μ L of solution B (12 mL of NaOH 1M mixed with 14.4 mL of distilled water) was added without further incubation. From |
| 311312313 | AlCl ₃ were added and left to rest for 1 min. Finally, 440 μ L of solution B (12 mL of NaOH 1M mixed with 14.4 mL of distilled water) was added without further incubation. From this reaction, 150 μ L were transferred to a 96-well microplate in triplicate. The samples |
| 311312313314 | AlCl ₃ were added and left to rest for 1 min. Finally, 440 μ L of solution B (12 mL of NaOH 1M mixed with 14.4 mL of distilled water) was added without further incubation. From this reaction, 150 μ L were transferred to a 96-well microplate in triplicate. The samples were read at 496 nm in a microplate reader (Synergy H1, BioTek, Vermont, USA). The |
| 311 312 313 314 315 | AlCl ₃ were added and left to rest for 1 min. Finally, 440 μ L of solution B (12 mL of NaOH 1M mixed with 14.4 mL of distilled water) was added without further incubation. From this reaction, 150 μ L were transferred to a 96-well microplate in triplicate. The samples were read at 496 nm in a microplate reader (Synergy H1, BioTek, Vermont, USA). The absorbance was compared with a Catechin standard curve ranging from 0.1 to 1 mg/ml. |
| 311 312 313 314 315 316 | AlCl ₃ were added and left to rest for 1 min. Finally, 440 µL of solution B (12 mL of NaOH 1M mixed with 14.4 mL of distilled water) was added without further incubation. From this reaction, 150 µL were transferred to a 96-well microplate in triplicate. The samples were read at 496 nm in a microplate reader (Synergy H1, BioTek, Vermont, USA). The absorbance was compared with a Catechin standard curve ranging from 0.1 to 1 mg/ml. The results were expressed as milligrams of Catechin equivalents (CE) per gram of dried |

318 **2.7** Antioxidant Capacity assessment by ABTS and FRAP methods

319 The total antioxidant activity of all samples was measured by ABTS (2,2'-Azino-bis(3-320 ethylbenzothiazoline-6-sulfonic acid)) assay (Re et al., 1999) with some modifications. 321 The ABTS+ stock solution was prepared by mixing 5 ml of 7 mM ABTS solution and 322 88 μ l of 140 mM potassium persulfate (K₂S₂O₈) solution. Then, the mixture was kept in 323 the dark and at room temperature for at least 16 h before use. The working solution of 324 ABTS+ was obtained by diluting the ABTS+ stock solution with distilled water to an 325 absorbance of 0.70 ± 0.02 at 734 nm. Then, 5 µl of the sample was added to 245 µl of 326 ABTS++ working solution, and the mixture was homogenised and then incubated in the 327 dark for 5 min. The absorbance of the control and the samples were recorded at 734 nm 328 using a microplate reader (Synergy H1, BioTek, Vermont, USA). The scavenging activity 329 of each sample on ABTS++ was calculated from a Trolox standard curve at concentrations 330 of 0.04 to 0.4 mg/mL. Results were expressed as micromole Trolox equivalents (TE) per 331 gram of dry extract.

For the Ferric Reducing Antioxidant Power (FRAP) assay (Benzie & Strain, 1996), 10 µl
of the sample was added to 300 µl of FRAP reagent in a microcentrifuge tube and
vortexed for 10s. Then, in triplicate, 100 µl of this mixture was transferred into a 96-well
microplate, and absorbance was measured at 595 nm in a microplate reader (Synergy H1,
BioTek, Vermont, USA). An ascorbic acid standard curve from 0.01 to 0.2 mg/mL was
used for the quantification. Results were expressed as micromole ascorbic acid
equivalents (AAE) per gram of dry extract.

339

341 **2.8 Statistical Analysis**

The data were subjected to a One-Way ANOVA using IBM[®] SPSS[®] Statistics 27 software, where statistical differences were noted. Differences among different treatments were determined using independent samples t-test for particle size and gastrointestinal results. For the metabolic activity, differences were determined by Dunnett's multiple comparison test, as this is more suitable for the mean comparison of different experimental groups against a control group. The significance level was defined at *p*<0.05, and the results are reported as means \pm SD.

349

350 3. Results and discussion

351 **3.1 Characterization of grape marc extract**

352 Hydroalcoholic extractions have proven to be efficient for extracting phenolics from 353 grape by-products (MohdMaidin et al., 2018, 2019; Spigno et al., 2007, 2017). Indeed, 354 we obtained a phenolics-rich extract with high total phenolic content (TPC), total 355 flavonoid content (TFC) content, and antioxidant capacity (Table 1). The phenolics 356 content was higher than those reported by Pintać et al. (2018) and Aresta et al. (2020). 357 They obtained 69 and 70 mg gallic acid equivalents (GAE)/g de, respectively, when 358 conventional extraction of polyphenols from grape marc. However, we obtained a lower 359 content of total monomeric anthocyanin content (TMAC), which might be explained by 360 a combination of factors such as extraction method, grape variety, growing region, and 361 processing, these conditions play a significant role since not all grapes bear the same 362 TMAC (Rinaldi et al., 2020; Spigno et al., 2015).

364 **3.2 Biocompatibility of grape marc extract**

365 Studying the potentially toxic effects of bioactive compounds is essential to determine 366 whether they are safe to consume without harming the host. The grape marc extract 367 (GME) showed a dose-responsive effect after 24 h of incubation with Caco-2 cells (Fig. 368 1). We observed cellular compatibility, *i.e.*, more than 70% of cell viability, for 33 and 369 67 µg/mL TPC based on GAE. However, cell viability below 70% was observed at the 370 highest concentration tested (100 GAE µg/mL), which is considered toxic. Studies in the 371 grape phenolic extract have shown that concentrations between 0.1 to 10 µg/mL present no toxicity in Caco-2 cells with up to 93% viability (Wang et al., 2016). Another study 372 373 by Costa et al. (2019) showed that concentrations of up to 2% of GME were non-toxic 374 for Caco-2 cells before and after simulated in vitro digestion. Also, Wolfe et al. (2008) 375 observed that concentrations below 60 mg/mL of different extracts, e.g., wild blueberry, 376 red grape, and strawberry, showed no cytotoxicity in HepG2 cells. However, in a 377 preliminary assay, we observed that concentrations of 5 mg/mL GME, in the 378 concentration range of some reports, were highly toxic (0% viability) for Caco-2 cells 379 (data not shown), highlighting the importance of assessing each extract for its safe 380 application.

381

382 **3.3 Cellular antioxidant activity (CAA) of grape marc extract**

Reactive oxygen species (ROS) are natural by-products of cell activity and essential signaling molecules (Zhang et al., 2016). However, an imbalance between oxidantproducing systems and antioxidant defense mechanisms can trigger cell damage and cause cell death (Alfadda & Sallam, 2012). Cell-based assays have been used to assess the effectiveness of dietary antioxidant compounds (Kellett et al., 2018). 388 Studies of intracellular oxidant production in Caco-2 cells were evaluated using 2'-7'-389 Dichlorodihydrofluorescein (DCFH) fluorescence, testing GME at non-toxic 390 concentrations (33 and 67 GAE µg/mL based on TPC). As shown in Fig. 2A, both GME 391 concentrations decreased the intracellular ROS basal levels, comparing with the control 392 (cells treated with Hanks' balanced salt solution (HBSS)) to a similar level to the one 393 observed for (±)-6-Hydroxy-2,5,7,8-tetramethylchromane-2-carboxylic acid (Trolox, 50 394 µg/mL). This result suggests that GME can reduce ROS naturally produced by the Caco-395 2 cells, demonstrating a possible antioxidant effect (intracellular) against ROS.

396 To evaluate the potential protective effect of GME against intracellular oxidation, Caco-397 2 cells were pre-treated with GME at the non-toxic concentrations of 33 and 67 µg 398 GAE/mL based on TPC for 4 h. Then, cells were stimulated with 5 µM of the oxidising 399 agent 3-Morpholinosydnonimine (Sin-1), selected according to the literature (PD ISO/TS 400 19006:2016). Cells treated with HBSS and stressed with Sin-1 were used as a positive 401 control. As shown in Fig. 2B, cells pre-treated with non-toxic concentrations of GME 402 significantly reduced intracellular ROS level produced after stimulation with Sin-1 403 compared to cells pre-treated with HBSS (control). This reduction was similar to that 404 observed for treated cells with 50 µg/mL Trolox which was used as a potent antioxidant 405 model compound.

GME showed a similar antioxidant effect to a well-known compound at similar concentrations, suggesting that GME polyphenols can effectively neutralise ROSinduced production (protective effect) in Caco-2 cells, demonstrating intracellular antioxidant capacity. The results of the CAA also corroborate the high antioxidant capacity of the GME observed by 2,2'-Azino-bis(3-ethylbenzothiazoline-6-sulfonic acid (ABTS) and ferric reducing antioxidant power (FRAP) methods. Wang et al. (2016)

412 induced ROS production using t-BOOH (tert-butyl hydroperoxide) in Caco-2 cells treated 413 with grape phenolic extract for 1 h, and their results showed that concentrations of 0.1 to 414 10 µg/mL exert an antioxidant effect over ROS. Other studies have reported that 415 concentrations of 100µ/mL, 200µ/mL and 500µ/mL reduced ROS production in Caco-2 416 cells treated with grape pomace extract for 5 h (Martins et al., 2017, 2020). However, at 417 500 µg/mL, the production of ROS was significantly reduced due to the pro-oxidant effect 418 of polyphenols (Martins et al., 2020). Milinčić et al. (2021) observed an EC50 of ABAP 419 (2,2'-azobis(2-amidopropane)) radical at a 54 mg TPC/mL concentration of grape pomace 420 skin extract on the same cell line. The concentrations used in the previously mentioned 421 studies are considerably higher than the ones we reported, indicating that while grape 422 pomace is an excellent source of antioxidants, the analyses of cell biocompatibility and 423 antioxidant capacity need to be carried out before their formulation as nutraceuticals or 424 functional food ingredients.

425

426 **3.4 Encapsulated GME morphology, size, and Z-potential**

427 The morphology and size of the encapsulated GME were studied through scanning 428 electron microscopy (SEM) analysis. Fig. 3A shows the formation of large crystals with 429 a wide distribution of submicron and micron particles during freeze-drying of GME 430 (Table 2). For the nano spray dried particles, different morphologies were observed for 431 the W-P particles with and without GME. Blank microparticles (W-P) had a spherical 432 shape and smooth surface (Fig. 3B), while microparticles loaded with GME (W-P-GME) 433 (Fig. 3C) kept their spherical shape but presented some wrinkles in their surface. 434 Moreover, no breakage was seen in W-P and W-P-GME. Regarding the size, W-P-GME 435 particles showed a smaller and narrower size distribution than W-P particles (Table 2).

436 Studies on the encapsulation of raw grape marc extract by conventional spray drying have 437 reported sizes of 9.8 µm when using pectin and casein, and 15 µm when using whey 438 protein isolate (WPI) alone. (Carra et al., 2022; Moreno et al., 2018). The results obtained 439 here (1µm) demonstrate that nano spray drying significantly affects the particles' 440 reduction size. Moreover, the particles we obtained displayed a more homogeneous and 441 well-defined particle shape than those in previously mentioned studies, where irregular 442 and dented surfaces were obtained, and in the case of WPI, holes were seen in the 443 microparticles (Moreno et al., 2018). The zeta potential of W-P-GME (Table 2) showed 444 a medium to high particle surface charge, which confers the particles' colloidal stability.

445

446 **3.5 Fourier Transform Infrared (FTIR) analysis**

447 FTIR analysis was used to examine interactions between the biopolymers and GME. The 448 infrared spectra of the carriers, GME and microparticles are shown in Fig. 4. For WPI, 449 characteristic amide I and II bands can provide information about protein secondary 450 structures, and their change in vibration frequencies is related to the interaction between 451 their functional groups. Amide I, represents the C=O carbonyl stretching vibration of the 452 peptide backbone (1600-1700 cm⁻¹), and the amide II band (<1550 cm⁻¹) represents the 453 C-N stretching and N-H bending (López-Rubio & Lagaron, 2012; Meng et al., 2021). As 454 for the GME, the characteristic bands of grape phenolic compounds were observed between 1700 and 900 cm⁻¹. The band at 1710 cm⁻¹ was attributed to the stretching in the 455 456 carbonyl group (C=O) band, 1600 and 1510 cm⁻¹ bands correspond to the C=C stretching, characteristic of aromatic systems. The peak around 1440 cm⁻¹ corresponds to the 457 458 antisymmetric in-plane bending of -CH₃ related to aromatic rings and flavonoids (Moreno 459 et al., 2018; Zhao et al., 2015). Characteristic peaks of pectin can be observed at 2920, 460 1740, 1610 and 900-1250 cm⁻¹ corresponding to the C-H stretching of the CH, CH2 and
461 CH3 groups, C=O stretching vibration of the ester carbonyl, C=O stretching of the
462 vibration the carbonyl group, C-O-C and O-H of pyranose rings respectively (Khodaiyan
463 & Parastouei, 2020).

464 Looking at the infrared spectra of W-P and W-P-GME, slight shifts in the amide I and 465 amide II regions were observed compared to WPI (1517 to 1535 cm⁻¹). These shifts can 466 be attributed to the interaction between carboxyl groups of pectin and the charged amino 467 groups of the main WPI proteins' composition (beta-lactoglobulin, alpha-lactalbumin, 468 and serum albumin) (Raei et al., 2018). An increase in the intensity was observed for the 469 W-P particles, which can be attributed to the rise in random coils and the previously 470 mentioned interaction between WPI and pectin (El-Messery et al., 2020; He et al., 2016). 471 However, when GME is added, a decrease in intensity is observed. This result is 472 consistent with those obtained by Meng and Li (2021), where Gallic acid, chlorogenic 473 acid, and epigallocatechin gallate-WPI complexes showed decreased intensity in the 474 amide I band. This change can be attributed to the reduction of α -helical structures as a 475 result of protein conformational modifications upon phenolics complexation by hydrogen 476 bonding and hydrophobic interactions between the phenolic compounds and hydrophobic 477 groups of the protein, so there are not only interactions but also changes in the secondary 478 structure of the proteins (Bourassa et al., 2013; He et al., 2016). According to previous 479 reports, W-P-GME did not show any characteristic band from GME, indicating that 480 phenolics distinct peaks can be hidden when in contact with other biopolymers like WPI. 481 This change could mean the formation of complexes that reduce the bending and 482 stretching of the bonds in GME polyphenols.

483 **3.6** *In vitro* digestion of free and encapsulated GME

484 The results of the residual TPC and antioxidant activity for both free and encapsulated 485 GME are shown in Fig. 5. These results represent the fraction of TPC (or activity which, 486 is quantified as Trolox equivalents (TE) or ascorbic acid equivalents (AAE)) remaining 487 after gastric digestion (GD) or gastrointestinal digestion (GID), the latter indicating the 488 bioaccessible fraction. Therefore, these values show the fraction of TPC (or activity) that 489 resisted the simulated gastrointestinal conditions in free GME. In contrast, for W-P-GME, 490 these values account for the fraction of TPC that resisted the conditions and/or was 491 encapsulated and effectively released from the microcapsules during digestion.

492 A different behaviour was observed for free and encapsulated GME, suggesting the 493 microcapsules play an essential role in the phenolic content and their activity during digestion. For free GME, we observed that the TPC underwent some degradation due to 494 495 the gastric conditions (acidic pH), as shown by a 76% residual TPC content (24% 496 unaccounted for; Fig. 5A). The moderate stability of GME polyphenols to gastric 497 digestion agrees with previous studies (Li et al., 2023). The free GME suffered further 498 degradation after intestinal conditions, resulting in a further 30% TPC loss in relation to 499 that remaining after GD; low stability of polyphenols has been reported at neutral pH 500 conditions (Li et al., 2023). So, after GID, the overall bioaccessible TPC was 54%. In the 501 case of encapsulated GME, about 30% of TPC was unaccounted for after GD (Fig. 5A), 502 which may represent the fraction not released from the microparticles. Indeed, high 503 preservation of the TPC was expected during GD since strong electrostatic interactions 504 stabilise the WPI- pectin complex at acidic pHs (3.6-4.5) (Raei et al., 2017), which should 505 protect phenolics from degradation. However, some release of phenolics will still occur 506 as WPI is susceptible to enzymatic hydrolysis, but pectin should have a stabilising effect 507 in the system (Reichembach & Lúcia de Oliveira Petkowicz, 2021; Wusigale et al., 2020). 508 Yet, the released fraction can also undergo similar degradation as that observed for the

free extract (GME). Therefore, assuming the residual 70% TPC content in W-P-GME will undergo similar degradation as that of the free extract during GID, values close to 54 % of residual TPC (as in GME) would be expected however, it was found that 83% of the TPC remained after GID. This indicates a protective effect of the microcapsules, which resulted in about 30% of the TPC in the gastric phase and their release at intestinal conditions, with an overall increase in the remaining TPC compared to free GME.

515 The behaviour of antioxidant activity during GID for both free and encapsulated GME 516 showed a similar trend to TPC. Thus, the free GME showed a slight loss of activity after 517 GD followed by a more pronounced decrease after GID, while for W-P-GME, the activity 518 was slightly increased after GID compared to GD (Fig. 5B & 5C). Besides, free GME's 519 bioactivity directly correlates with residual TPC values after GD and GID, achieving 520 values of 73% and 57% of the initial activity, as assessed by the ABTS method. Although 521 a similar trend was observed in both phases, lower values were recorded using the FRAP 522 method.

523 For encapsulated GME, although a positive correlation was observed between residual 524 TPC and antioxidant activity, the latter showed lower values than the residual TPC. For 525 instance, 29 and 61% of the activity was observed using the ABTS method after GD and 526 GID in W-P-GME. The reduced activity compared to the residual TPC might be due to 527 released polyphenols from the capsules bearing lower antioxidant activity than those that 528 were still encapsulated or that they might be complexed with the capsule components 529 since they are known to interact with whey proteins and their peptides (Guo & Jauregi, 530 2018), which has been confirmed by the FTIR spectra.

531 Overall, the results of GID showed that the encapsulation succeeded in preserving the532 TPC and increasing their bioaccessibility. For the antioxidant activity, similar results to

free GME were observed according to the ABTS method, and slightly higher activityaccording to the FRAP method.

535

536 4. Conclusions

537 A raw ethanolic extract of a winery by-product (grape marc) with antioxidant capacity 538 was successfully encapsulated using whey protein isolate (WPI) and pectin and nano 539 spray drying (73% yield), resulting in spherical smoothed-surface microparticles with an 540 average size of 1 µm, polydispersity index (PDI) of 0.717, and a surface charge (Z-541 potential) close to -30 mV. The Fourier Fourier Transform Infrared (FTIR) analysis of 542 the microparticles confirmed the complexations between WPI, pectin and the phenolics 543 in grape marc extract (GME) through non-covalent interactions. The developed 544 encapsulation system protected the GME phenolics and the antioxidant activity during 545 gastrointestinal digestion (GID), improving bioaccessibility. The potent antioxidant 546 intracellular protective effect of GME observed, and its improved resistance to GID when 547 encapsulated compared to the free form suggest this encapsulation system could be a 548 promising strategy towards preserving the antioxidant activity of this high-value-added 549 by-product of the wine industry. The selected wall materials proved that the 550 microcapsules resisted gastric conditions and could provide a targeted release in the lower 551 intestine, where phenolic compounds are absorbed and can be metabolised by the 552 microbiota. Although further studies are needed to test the stability, biocompatibility, and 553 in vivo bioactivity of the WPI-pectin-GME microcapsules, the presented results are 554 promising towards using encapsulated GME as a nutraceutical.

555

556 **Conflict of Interest**

- 557 The authors declare that the research was conducted without any commercial or
- 558 *financial relationships that could be construed as a potential conflict of interest.*
- 559

560 Author Contributions

561 Data curation; Formal analysis; Methodology; Writing-original draft: Aimara V. De La

562 Cruz Molina. Conceptualisation: Isabel Rodriguez, & Lorenzo Pastrana. Funding

563 *acquisition:* Lorenzo Pastrana. *Resources:* Lorenzo Pastrana. *Writing-review & editing:*

- 564 Isabel Rodriguez, Aimara V. De La Cruz Molina, Catarina Gonçalves, Mafalda D. Neto
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