

Numerical investigations of reactive pollutant dispersion and personal exposure in 3D urban-like models

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3 Numerical investigations of reactive pollutant
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19

20 **Abstract**

21 With satisfactory validation by experimental data, we perform computational
22 fluid dynamic(CFD) simulations with the standard $k-\varepsilon$ model to investigate how
23 NO-NO₂-O₃ photochemistry and turbulent mixing influence reactive pollutant
24 dispersion and vehicular NO_x exposure in 21-row(neighborhood-scale~1km)
25 three-dimensional(3D) medium-dense urban models with an approaching wind
26 parallel(perpendicular) to the main(secondary) streets. Personal intake fraction P_{iF}
27 and its spatially-averaged values for the entire building (i.e. building intake fraction
28 $\langle P_{iF} \rangle_B$) are adopted for reactive/passive exposure analysis with/without
29 NO_x-O₃-photochemistry.

30 Some meaningful findings are proposed: 1) There are flow adjustment processes
31 coupling turbulent mixing and chemical reactions through urban areas(i.e. secondary
32 Street 1 to 20). NO-NO₂-O₃ photochemistry induces O₃ depletion and NO conversion
33 into NO₂ producing significant increase in NO₂ exposure and slight decrease in NO
34 exposure compared with passive dispersion. 2) With span-wise NO_x sources, Street 10
35 in the fully-developed region experiences weaker wind and subsequently greater
36 $\langle P_{iF} \rangle_B$ (0.207ppm) than Street 3(0.135ppm) in the upstream flow-adjustment region.
37 $\langle P_{iF} \rangle_B$ descends exponentially from the target building toward downstream, and
38 Street 10 experiences quicker decay rates. 3) With stream-wise NO_x sources along the
39 main street, $\langle P_{iF} \rangle_B$ first ascends, then reaches equilibrium values
40 (e.g.0.046-0.049ppm for passive). 4) If background O₃ concentration [O₃] rises from
41 20ppbv to 40 and 100ppbv, more NO is oxidized by O₃ to generate NO₂. As
42 [O₃]=20ppbv, if NO-NO₂ emission ratio decreases from 10 to 5, NO₂ exposure is

43 partly offset but NO exposure change little. Present methodologies are confirmed
44 effective to investigate impacts of more complicated meteorological conditions and
45 chemical mechanisms on exposure in urban districts.

46

47 **Keywords:** NO-NO₂-O₃ photochemistry; reactive pollutant dispersion; personal
48 intake fraction (P_{iF}); building intake fraction ($\langle P_{iF} \rangle_B$); computational fluid
49 dynamic (CFD) simulation; three-dimensional (3D) urban models

50

51 **1. Introduction**

52 Following the ongoing urbanization worldwide, vehicular pollutant emissions
53 have become one of the major sources in urban air pollution[1-3]. Heavy traffic flows,
54 compact urban configurations and unfavorable meteorological conditions are the main
55 reasons of large pollutant exposure and adverse health impacts on city dwellers[4]. On
56 average, people spend more than 90% of their time indoors. Outdoor air pollutants in
57 urban areas can penetrate into indoor via doors, windows, building cracks and other
58 ventilation duct systems[5-7]. Particularly, vehicular pollutant exposure to urban
59 residents living near busy roads should be paid more attention, because they suffer
60 from higher health risks than other urban microenvironments[7-10]. Apart from
61 reducing vehicular pollutant emissions, sustainable urban design with better
62 understanding the influence of urban layouts and atmospheric conditions on the flow
63 and passive/reactive pollutant dispersion in urban areas can help enhancing pollutant

64 dilution and mitigating traffic-related pollutant exposure[11-15].

65 As reviewed by the literature[16-29], in the past decades, a number of
66 computational fluid dynamic (CFD) simulations, outdoor field observations and wind
67 tunnel experiments have been widely performed to clarify turbulent flow
68 characteristics and pollutant dispersion in urban models from street-scale (~100m) to
69 neighborhood-scale (~1km). Street aspect ratios (H/W) are reported as the most key
70 urban parameters in two-dimensional (2D) street canyons[16-18, 25-26, 29-32].
71 Realistic urban districts are usually three-dimensional (3D) with pollutant exchange
72 across street roofs and lateral/stream-wise urban boundaries. Generally, the building
73 planar area index λ_p (i.e. the ratio between the planar area of buildings viewed from
74 above and the total floor area) and the frontal area index λ_f (i.e. the ratio of the frontal
75 area of buildings to the total floor area) are typical building packing density indexes
76 and key parameters of 3D urban areas[33-38]. Moreover, other significant factors
77 have been also verified, such as building height variations[38-40], ambient wind
78 directions[40-41], elevated building design[42-44], tree planting[46-49] etc. In
79 addition, thermal stratification and buoyancy forces induced by wall heating and solar
80 shading also significantly influence or dominate the flow and pollutant dispersion if
81 wind speed is relatively small and Richardson number is large[50-56].

82 Recently, several researches[7, 44, 57] introduce personal intake fraction (P_{iF})
83 to quantify street-scale pollutant exposure induced by vehicle emissions. In contrast to
84 population intake fraction (IF) for a specific population[4, 58], P_{iF} is independent of
85 population density and size which represents the fraction of pollutants inhaled

86 averagely by each person of a population to the total pollutant emissions. For example,
87 1ppm (part per million or 10^{-6}) means the inhalation of 1mg pollutants if 1kg
88 pollutants being emitted. By performing CFD simulations validated by experimental
89 data, [Hang et al.\[7\]](#) estimated spatial mean P_{iF} (i.e. $\langle P_{iF} \rangle \sim 1-5\text{ppm}$) of passive
90 pollutant (i.e. CO) in shallow 2D street canyons ($H/W=0.5-1$). Later, $\langle P_{iF} \rangle$ in 3D
91 urban district models were confirmed one order smaller ($\sim 0.1\text{ppm}$) than 2D models
92 with similar aspect ratios ($H/W=0.5-1$)[[41](#), [57](#)].

93 Besides the dynamic dispersion of passive pollutants, there are chemical
94 processes of reactive pollutants in urban streets, such as the $\text{NO}_x\text{-O}_3$ photochemistry
95 [[59-64](#)], $\text{NO}_x\text{-O}_3\text{-VOCs}$ chemical mechanisms[[65-69](#)] etc. Among them, the impacts
96 of different heating scenarios and building configurations on reactive dispersion are
97 extensively investigated through LES or RANS approaches, such as various shading
98 settings[[61](#)], wall heating scenarios[[60](#), [62](#)], aspect ratios[[63](#)] etc. However, most
99 studies so far mainly examine reactive pollutant dispersion in 2D street canyons while
100 investigations on reactive pollutant dispersion and the related exposure in 3D urban
101 models are still rare. Therefore, this study incorporates $\text{NO-NO}_2\text{-O}_3$ chemistry into
102 CFD simulations and numerically investigates reactive pollutant dispersion and
103 exposure in urban models. As a start, the impacts of various ground-level source
104 locations and reactant proportions ($\text{NO:NO}_2\text{:O}_3$) in 3D medium-dense urban models
105 ($H/W=1$, $\lambda_p=\lambda_f=0.25$) are studied under neutral meteorological conditions.

106 The sketch of this paper is organized as follows: Section [2](#) introduces the indexes
107 for pollutant dispersion and exposure. Section [3](#) illustrates model setups and all test

108 cases in CFD simulations. Section 4 presents the flow and pollutant dispersion
109 validations by wind tunnel data. Section 5 shows results and discussions, and Section
110 6 draws the conclusions.

111

112 2. Indexes for pollutant dispersion and exposure

113 2.1 Personal intake fraction (P_{iF}) and building intake fraction ($\langle P_{iF} \rangle_B$)

114 Population intake fraction (IF) and personal intake fraction (P_{iF}) are effective
115 indexes to quantify vehicular pollutant exposure in local streets or neighborhoods[41,
116 44, 57]. Both exposure indexes are defined as Eq. (1):

$$117 \quad IF = \sum_i^N \sum_j^M P_i \times Br_{i,j} \times \Delta t_{i,j} \times Ce_j / \dot{m} \quad (1a)$$

$$118 \quad P_{iF} = IF / \sum_j^M P_i \quad (1b)$$

119 where, N and M are the number of age groups and microenvironments, P_i represents
120 the population size for the age group i , $Br_{i,j}$ (m^3/s) and $\Delta t_{i,j}$ (s) are the average
121 breathing rate and the individual time spent for the i th age group in the j th
122 microenvironment, Ce_j (kg/m^3) denotes pollutant concentration in microenvironment j
123 and \dot{m} (kg) means total emissions released from vehicles. It is worth mentioning that
124 P_{iF} is independent of pollutant release rates as well as population density and size,
125 but can be influenced by building configurations, meteorological conditions and
126 pollutant source settings etc.

127 According to the literature[4, 70-71], the population data are categorized into

128 three subgroups($N=3$, Fig. 1a): Children(21.2%), Adults(63.3%) and Elders(15.5%).
 129 Moreover, the time activity patterns for each subgroup are divided into four
 130 microenvironments($M=4$, Fig. 1b): indoors at home($j=1$), other indoor locations($j=2$),
 131 in or near vehicles($j=3$), and other outdoor locations($j=4$). Table 1 lists activity time
 132 patterns and breathing rates in each subgroup for indoors at home($j=1$). In this study,
 133 all present building models are supposed to residential type and only $j=1$ (indoors at
 134 home) is adopted to calculate P_{iF} [44, 57]. Especially, the area-averaged P_{iF} of the
 135 building wall surface is denoted as wall intake fraction($\langle P_{iF} \rangle_w$). The
 136 spatially-averaged P_{iF} of the entire building surfaces is represented as building
 137 intake fraction($\langle P_{iF} \rangle_B$), which means the fraction of total traffic emissions inhaled
 138 averagely by each person living in this roadside building.

139

140 2.2 Photostationary state defect (d_{ps} , unit: %)

141 Referring to previous researches[59-60], the photostationary state defect (d_{ps}) is
 142 an effective indicator to measure the departure degree from photochemical
 143 equilibrium and can be expressed in the following form:

$$144 \quad d_{ps} = \left(\frac{k_I[\text{NO}][\text{O}_3]}{J_{\text{NO}_2}[\text{NO}_2]} - 1 \right) \times 100 \quad (2)$$

145 here, $k_I[\text{NO}][\text{O}_3]$ and $J_{\text{NO}_2}[\text{NO}_2]$ represent the depletion and generation rates of
 146 ozone. d_{ps} is positive value if ozone depletion rate greater than its formation rate, and
 147 vice versa. Especially, d_{ps} equals zero when chemical reactions reach equilibrium
 148 state.

149

150 3. CFD setups and case descriptions

151 3.1 Numerical approaches

152 With advances in computer technologies, CFD as a powerful modelling tool has
153 been widely employed to reproduce turbulent flow structure as well as to predict
154 pollutant dispersion and transport in urban districts. Though large eddy simulations
155 (LES) have been confirmed to be more accurate in predicting turbulence than
156 Reynolds-Averaged Navier-Stokes (RANS) models[27, 73-75], RANS approaches are
157 still extensively used, since LES models require expensive computational expenses
158 and have challenges in selecting sub-grid scale models and specifying appropriate
159 boundary conditions. Moreover, among the RANS models (e.g. various $k-\varepsilon$ and $k-\omega$
160 models), the standard $k-\varepsilon$ model shows good agreements with experimental data and
161 has been widely adopted[7-8, 37-38, 45, 57-58, 76-79], although it has limitations in
162 predicting turbulent kinetic energy in strong-wind regions. Therefore, by considering
163 model performance and computational loads, the standard $k-\varepsilon$ model is selected to
164 solve the steady-state isothermal flow field. The governing equations for the flow and
165 turbulent quantities implemented are as below[80]:

166 The mass continuity equation:

$$167 \frac{\partial \bar{u}_j}{\partial x_j} = 0 \quad (3)$$

168 The momentum equation:

169
$$\bar{u}_j \frac{\partial \bar{u}_i}{\partial x_j} = -\frac{1}{\rho} \frac{\partial \bar{p}}{\partial x_i} + \frac{\partial}{\partial x_j} \left(\nu \frac{\partial \bar{u}_i}{\partial x_j} - \overline{u_i' u_j'} \right) \quad (4)$$

170 The transport equations of turbulent kinetic energy (k) and its dissipation rate (ε):

171
$$\bar{u}_i \frac{\partial k}{\partial x_i} = \frac{\partial}{\partial x_i} \left[\left(\nu + \frac{\nu_t}{\sigma_k} \right) \frac{\partial k}{\partial x_i} \right] + P_k - \varepsilon \quad (5)$$

172
$$\bar{u}_i \frac{\partial \varepsilon}{\partial x_i} = \frac{\partial}{\partial x_i} \left[\left(\nu + \frac{\nu_t}{\sigma_\varepsilon} \right) \frac{\partial \varepsilon}{\partial x_i} \right] + C_{\varepsilon 1} \frac{\varepsilon}{k} P_k - C_{\varepsilon 2} \frac{\varepsilon^2}{k} \quad (6)$$

173 where, \bar{u}_j is mean velocity components ($\bar{u}_j = \bar{u}, \bar{v}, \bar{w}$ as $j=1, 2, 3$); ν and $\nu_t = C_\mu \frac{k^2}{\varepsilon}$

174 ($C_\mu=0.09$) represent the kinematic viscosity and the eddy viscosity, respectively; the

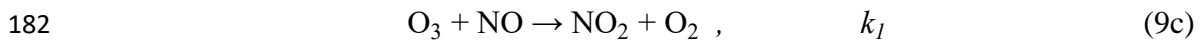
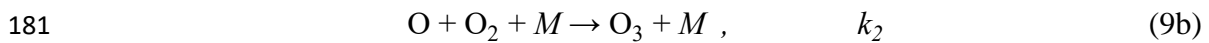
175 Reynolds stress tensor $-\overline{u_i' u_j'}$ and the turbulence production term P_k are defined as:

176
$$-\overline{u_i' u_j'} = \nu_t \left(\frac{\partial \bar{u}_i}{\partial x_j} + \frac{\partial \bar{u}_j}{\partial x_i} \right) - \frac{2}{3} k \delta_{ij} \quad (7)$$

177
$$P_k = \nu_t \times \frac{\partial \bar{u}_i}{\partial x_j} \left(\frac{\partial \bar{u}_i}{\partial x_j} + \frac{\partial \bar{u}_j}{\partial x_i} \right) \quad (8)$$

178 Note that δ_{ij} is the Kronecker delta whose value is 1 when $i=j$ and otherwise is 0.

179 The NO_x-O₃ simple photochemical mechanism is described as follows[59-60]:



183 Here, M denotes a third molecule, for example O₂ or N₂, to absorb excess energy and

184 stabilize O₃ molecule formed; J_{NO_2} , k_2 and k_1 represent rate constant for each

185 reaction, respectively. Since the oxygen atom (O) is highly reactive, it combines

186 rapidly with O₂ once O is produced from NO₂ photolysis. This is so-called

187 pseudo-steady-state approximation that the depletion and production rates of O₃ are

188 nearly equal[81]:

$$189 \quad k_2[\text{O}][\text{O}_2][\text{M}] = J_{\text{NO}_2}[\text{NO}_2] \quad (10)$$

190 Therefore, Eq. (9a) is the rate control step for O₃ formation. Based on the above
 191 assumption, the transport equations for reactive pollutants can be defined as[59-60]:

$$192 \quad \bar{u}_j \frac{\partial[\text{NO}]}{\partial x_j} = D_m \frac{\partial^2[\text{NO}]}{\partial x_j \partial x_j} + \frac{\partial}{\partial x_j} \left(D_e \frac{\partial[\text{NO}]}{\partial x_j} \right) + J_{\text{NO}_2}[\text{NO}_2] - k_I[\text{O}_3][\text{NO}] + S_{\text{NO}} \quad (11a)$$

$$193 \quad \bar{u}_j \frac{\partial[\text{NO}_2]}{\partial x_j} = D_m \frac{\partial^2[\text{NO}_2]}{\partial x_j \partial x_j} + \frac{\partial}{\partial x_j} \left(D_e \frac{\partial[\text{NO}_2]}{\partial x_j} \right) - J_{\text{NO}_2}[\text{NO}_2] + k_I[\text{O}_3][\text{NO}] + S_{\text{NO}_2} \quad (11b)$$

$$194 \quad \bar{u}_j \frac{\partial[\text{O}_3]}{\partial x_j} = D_m \frac{\partial^2[\text{O}_3]}{\partial x_j \partial x_j} + \frac{\partial}{\partial x_j} \left(D_e \frac{\partial[\text{O}_3]}{\partial x_j} \right) + J_{\text{NO}_2}[\text{NO}_2] - k_I[\text{O}_3][\text{NO}] \quad (11c)$$

195 where, D_m and D_e are the molecular and eddy diffusivity; the third terms on the
 196 right-hand side of Eq. (11) represent chemical reaction term; S_{NO} and S_{NO_2} denote the
 197 source terms of NO and NO₂; the Schmidt number $Sc_t = \nu_t / D_e$ is specified as 0.7[33, 39,
 198 46, 48, 78]. Furthermore, the photolysis rate J_{NO_2} and rate constant k_I are calculated by
 199 [59-60]:

$$200 \quad J_{\text{NO}_2} = 8.14 \times 10^{-3} \left\{ 0.97694 + 8.3700 \times 10^{-4} (T - 273.15) + 4.5173 \times 10^{-6} \times (T - 273.15)^2 \right\} \quad (12a)$$

201)

$$202 \quad k_I = 4.405 \times 10^{-2} \exp\left(-\frac{1370}{T}\right) \quad (12b)$$

203 Here, T is temperature in K; the units of J_{NO_2} and k_I are s⁻¹ and ppbv⁻¹s⁻¹.

204 Additionally, the temperature-dependent rate constant is not considered in isothermal

205 flow of this study, thus J_{NO_2} and k_I are 8.1×10^{-3} s⁻¹ and 4.450×10^{-4} ppbv⁻¹s⁻¹ by

206 assuming reactive pollutants undergo chemical processes under the isothermal

207 condition with a fixed T of 298.15K.

208 All governing equations(Eqs. (3-6, 11)) are discretized by a finite volume
209 method (FVM) with the second-order upwind scheme. The SIMPLE algorithm is
210 employed for pressure and velocity coupling. The under-relaxation factors for
211 pressure, momentum, k and ε terms are specified as 0.3, 0.7, 0.8 and 0.8. Numerical
212 simulation does not stop until the absolute residuals of all variables are less than 10^{-6} .

213

214 **3.2 Three-dimensional (3D) urban model setups in CFD**

215 The building layouts, as depicted in Fig. 2a, are based on the idealized 3D
216 medium-dense urban clusters (i.e. street aspect ratio $H/W=1$; building packing density
217 $\lambda_p=\lambda_f=0.25$). To better illustrate model configurations, x , y and z are described as the
218 stream-wise, span-wise and vertical directions, respectively. The cubic building
219 models($H=B=30\text{m}$) with a uniform spacing($W_s=W_m=30\text{m}$) are constructed in x and y
220 directions. Moreover, the approaching wind is parallel to the main streets(x direction)
221 and perpendicular to the secondary streets(y direction). $x/H=0$ means the cross section
222 in windward street opening and $y/H=0$ denotes the central cross section of the main
223 street (Fig. 2c).

224 As verified by the literature[38-40, 82-83], the airflow in the middle column is
225 hardly affected by lateral boundaries if the lateral width of the urban model is
226 sufficiently large. Hence, only half of the middle column with two lateral symmetry
227 boundaries are considered during CFD simulations to reduce computational efforts
228 (Fig. 2b). Following the CFD guidelines[84-86], the distances between urban

229 boundaries and the domain top, domain inlet, domain outlet are $9H$, $6.7H$ and $32.3H$,
 230 respectively. Furthermore, the zero normal gradient boundary condition is adopted at
 231 two lateral boundaries and the domain top (i.e. symmetry) and the domain outlet (i.e.
 232 outflow). Several recent studies[5-6, 8-9, 44, 57-58] reported that, pollutant
 233 concentration on the wall surfaces of near-road buildings can be treated as indoor
 234 concentration originated from outdoor pollutants since the indoor/outdoor (I/O) ratio
 235 of pollutant concentration is nearly one for naturally-ventilated buildings[5, 72].
 236 Therefore, by assuming all present building models are naturally-ventilated type, the
 237 flow and pollutant dispersion within indoor space of buildings are not taken into
 238 account to reduce the grid numbers and computational costs in CFD simulations. The
 239 literature have applied such technique to effectively quantify vehicular pollutant
 240 exposure in 2D street canyons or 3D urban models[5-6, 8-9, 44, 57-58].

241 For the domain inlet, the power-law time-averaged velocity profile $U_0(z)$ in the
 242 upstream free flow is adopted (Eq. 13a)[40-41, 57, 82-83] which is scaled to that in
 243 wind tunnel experiments[87]. Base on the CFD guideline[84-86], vertical profiles of
 244 $k(z)$ and $\varepsilon(z)$ are given by Eqs. (13b-c):

$$245 \quad U_0(z) = U_{ref} \times (z/H)^{0.16} \quad (13a)$$

$$246 \quad k(z) = u_*^2 / \sqrt{C_\mu} \quad (13b)$$

$$247 \quad \varepsilon(z) = C_\mu^{3/4} k^{3/2} / (\kappa_\nu z) \quad (13c)$$

248 where, U_{ref} is the reference velocity at the building height in upstream free flow
 249 ($U_{ref}=3.0$ m/s at $z = H$); the friction velocity $u_*=0.24$ m/s[40-41, 57, 82-83]; von

250 Karman constant $\kappa_v=0.41$, $C_\mu=0.09$. Furthermore, vertical profiles of Eq. (13)
251 represent neutral atmospheric boundary layer with a full-scale surface roughness
252 $z_0=0.1\text{m}$ [93] and have been adopted in previous studies[40-41, 57, 82-83].

253 In addition, Fig. 2c demonstrates the overhead and lateral views of mesh
254 distribution for test cases. The minimum grid size of 0.2 m near the wall surfaces and
255 the stretching ratio between adjacent grids of 1.15 (about 3.2 million hexahedral cells)
256 are applied to ensure sufficiently fine grid at the pedestrian level (0-1.5m) and near
257 building surfaces. As the normalized distance y^+ ($y^+=yu_\tau/\nu$) ranges from 30 to 600 at
258 most regions of wall surfaces, standard wall function with no slip boundary condition
259 is set on all wall surfaces. According to the literature[13, 40, 88-89], a specific
260 roughness modification is assigned to the upstream and downstream ground to obtain
261 a horizontally homogeneous ABL surrounding urban regions. Especially, the grid
262 independence tests are presented later in subsection 4.1.

263

264 3.3 Pollutant source settings and model description of test cases

265 Three kinds of pollutant source locations are considered in this study, i.e. the
266 span-wise (y direction) emission sources in the 3rd or 10th secondary street is denoted
267 as S3 and S10 (Fig. 3a-b), and the stream-wise (x direction) emission sources along
268 the main street is represented by Sm (Fig. 3c). The reactive pollutants involved in
269 present photochemistry are NO, NO₂ and O₃. Among them, NO_x is assumed to be
270 emitted from vehicles into the street canyon while background O₃ concentration is

271 specified at the domain inlet and entrained by approaching wind into urban districts.
272 The background concentrations for NO and NO₂ at the domain inlet are zero in this
273 study. Following the literature[59-60], typical automobile emission ratio of NO to
274 NO₂ ($R_{\text{NO}/\text{NO}_2}$) of 10 is adopted. NO and NO₂ are released from the lowest grid cell
275 ($z=0-0.2\text{m}$) at rates of 100 and 10 ppbv s⁻¹, which corresponds to emission strengths
276 of 1.227×10^{-7} and 1.88×10^{-8} kg m⁻³s⁻¹ ($P=1$ atm, $T=298.15\text{K}$). For S3 or S10 sources,
277 as the street width (W_s) is 30m, this NO_x emission intensity of 849 μg m⁻¹s⁻¹ is
278 equivalent to a traffic volume of about 6100 vehicles per hour when considering a
279 NO_x emission of 0.5g km⁻¹ per vehicle[59-60]. Particularly, we are mainly concerned
280 with the proportion among NO and NO₂ rather than the assigned emission rate for
281 each vehicular pollutant.

282 Furthermore, four O₃ background concentrations (i.e. [O₃] =1, 20, 40 and 100
283 ppbv) with $R_{\text{NO}/\text{NO}_2}=100:10$ are investigated to study the effects of different [O₃] on
284 reactive dispersion. In addition, Carslaw[90] verified that there is an increasing
285 NO₂/NO_x emission ratio in road traffic emissions referring to observation in real cities.
286 Thus, three emission ratios of NO to NO₂ (i.e. $R_{\text{NO}/\text{NO}_2}=100:10$, 50:10 and 100:20)
287 with [O₃]=20ppbv are considered to examine the impacts of various $R_{\text{NO}/\text{NO}_2}$ on
288 reactive pollutant dispersion.

289 Overall, total 11 test cases are described as Case P [Source, $R_{\text{NO}/\text{NO}_2}$] or Case R
290 [Source, $R_{\text{NO}/\text{NO}_2}$, [O₃]] and summarized in Table 2. Here ‘P’ means passive dispersion
291 without chemical reactions, ‘R’ denotes reactive dispersion with NO-NO₂-O₃
292 chemistry; ‘Source’ contains three pollutant source locations, i.e. S3, S10 and Sm,

293 respectively; $R_{\text{NO}/\text{NO}_2}$ represents the emission ratios of NO to NO₂; [O₃] is
294 background O₃ concentration (mole fraction, unit: ppbv).

295

296 **4 Validation study of flow and pollutant dispersion in 3D urban models**

297 **4.1 Flow validation by wind tunnel data**

298 In this subsection, the performances of various steady k - ε models (standard,
299 RNG and Realizable) with standard wall function are evaluated by wind tunnel data.
300 Moreover, the grid independence tests are also implemented.

301 As shown in Fig. 4a-b, idealized 3D urban models in wind tunnel experiment[87]
302 consist of 7 rows and 11 columns of regularly-aligned cubic buildings
303 ($H=B=W=15\text{cm}$). Vertical profiles of velocity components (\bar{u} , \bar{w}) and turbulence
304 kinetic energy (k) are measured at points of V_i ($i=1-6$), which are central positions in
305 the i th secondary street located at $y/H=1$ and $x/H=1.5H, 3.5H, 5.5H, 7.5H, 9.5H,$
306 $11.5H$ respectively (Fig. 4a-b).

307 In CFD validation study, the similar full-scale model configurations
308 ($H=B=W=30\text{m}$, Fig. 4c) are reconstructed with the scale ratio of 200:1 to
309 wind-tunnel-scale models. Moreover, all CFD setups including computational
310 domains, boundary conditions and convergence criteria in this full-scale CFD
311 validation study are similar with those in subsection 3.2 except that the distance
312 between urban boundaries and domain outlet is $40.3H$ (Fig. 4c). Based on the building
313 height($H=0.15\text{m}$ or 30m) and reference velocity($U_{ref}=3\text{m/s}$), the reference Reynolds

314 number $Re = \frac{U_{ref}H}{\nu} \approx 30000$ and 6×10^6 for wind-tunnel-scale and full-scale urban
315 models, which are much larger than 11000 satisfying Reynolds number independence
316 requirement. Furthermore, Fig. 4d depicts the coarse, medium and fine grid
317 arrangements with hexahedral cells of about 1, 2 and 3.3 million and the minimum
318 grid size of 0.4m, 0.2m and 0.1m, respectively.

319 Fig. 5 first shows results of grid independence test at V1(Fig. 5a-b), then depicts
320 vertical profiles of time-averaged stream-wise velocity $\bar{u}(z)$ at Points V1, V4 and V6
321 (Fig. 5c-e), vertical velocity $\bar{w}(z)$ (Fig. 5f) and turbulence kinetic energy $k(z)$ (Fig. 5g)
322 at Point V1 between numerical results and experimental data. There is little difference
323 in numerical results between the coarse, medium and fine grid, and thus medium grid
324 is selected for case studies to reduce computational loads. Besides, the standard $k-\varepsilon$
325 model with the medium grid shows better agreements with wind tunnel data than the
326 RNG and Realizable $k-\varepsilon$ models.

327 To further quantify the modeling accuracy and reliability of the standard $k-\varepsilon$
328 model with the medium grid arrangement, several statistical performance metrics are
329 applied, including the mean value, the standard deviation, the factor 2 (FAC2), the
330 normalized mean square error (NMSE), the fraction bias (FB) and the correlation
331 coefficient (R)[91]. Among which, the closer NMSE value to zero, the smaller
332 difference between experiment data and CFD results; FAC2 value larger than one
333 means over-prediction against experiment data while smaller than one represents
334 under-prediction; Similarly, the negative and positive FB values denote
335 overestimation and underestimation. Results of $\bar{u}(z)$ at Points V1, V4 and V6 as well

336 as $\bar{w}(z)$ and $k(z)$ at Point V1 are listed in [Table 3](#). Referring to COST Action 732's
337 recommended criteria[91], a credible CFD model should satisfy the following
338 statistical metrics standards: $0.5 \leq \text{FAC2} \leq 2$, $\text{NMSE} \leq 1.5$ and $-0.3 \leq \text{FB} \leq 0.3$.
339 Overall, metrics lie in the recommended criteria, particularly values of $\bar{u}(z)$ meet
340 well. However, values of $\bar{w}(z)$ and $k(z)$ reveal relatively poorer performance than
341 those of $\bar{u}(z)$, which is largely attributed to the limitation of the standard $k-\varepsilon$ model.
342 Conclusively, the validation study shows that present CFD methodologies applying
343 standard $k-\varepsilon$ model with medium grid have credible numerical accuracy in predicting
344 urban turbulent flow and can be employed for further case studies.

345

346 **4.2 Validation of pollutant dispersion by wind tunnel data**

347 Experimental and numerical studies have been extensively performed to
348 investigate passive pollutant dispersion in the idealized street canyons. Unfortunately,
349 there are currently little experimental data to directly validate the present CFD model
350 coupled with chemistry[59-60, 63-64]. However, the reactive pollutants considered in
351 this paper are effectively passive and can be divided into two subsets, i.e. total
352 nitrogen oxide ($\text{NO}_x = \text{NO} + \text{NO}_2$) and total oxidant ($\text{OX} = \text{NO}_2 + \text{O}_3$), since reactions ([Eq.](#)
353 [9](#)) interconvert NO with NO_2 , and O_3 with NO_2 but without redundant productions. In
354 addition, the chemical reaction terms cancel out when [Eqs. \(11a-b\)](#) and [Eqs. \(11b-c\)](#)
355 are added, which indicates that the transport and dispersion of NO_x and OX can be
356 deemed as passive scalar. Consequently, in this subsection, Standard $k-\varepsilon$ model with

357 standard wall function has been implemented and validated against wind tunnel
358 experiment to evaluate the reliability of numerical simulation in predicting passive
359 pollutant distribution.

360 The configurations of the wind tunnel measurement[92], as depicted in Fig.6a-b,
361 are consisted of nine rectangular building models ($L_x=27.6\text{cm}$, $L_y=18.4\text{cm}$, $H=8\text{cm}$)
362 with three rows and three columns (3×3) and uniform street widths ($W=8\text{cm}$, $H/W=1$).
363 Moreover, tracer gas(C_2H_6) is released from a line source (0.5cm in width, dx and
364 18.4 cm in length, L_s) at a velocity of $w_{source}=0.01$ m/s which is paralleled with y
365 direction and locates in the central street canyon in front of building No.2(Fig. 6a-b).
366 Furthermore, C_2H_6 concentration profiles are measured in the middle of leeward and
367 windward walls near the line source as well as on the central line of roof surface in
368 the building No.2(Fig. 6a-b).

369 In full-scale pollutant dispersion validation study, similar model configurations
370 ($L_x=138\text{m}$, $L_y=92\text{m}$, $H=40\text{m}$, in Fig. 6c) are rebuilt with the scale ratio of 500:1 to
371 wind-tunnel-scale models. CFD setups are similar with subsection 3.2, but the
372 distances between urban boundaries and the domain top, domain side, domain inlet,
373 domain outlet are $9H$, $5H$, $5H$ and $15H$, respectively. The approaching wind is parallel
374 to the main streets(x direction) and perpendicular to the secondary streets(y direction).
375 Furthermore, vertical profiles of stream-wise velocity $\bar{u}(z)$, turbulence kinetic energy
376 $k(z)$ and turbulent dissipation rate $\varepsilon(z)$ fitted by measured data in wind tunnel
377 experiments[92] are adopted at the domain inlet (Fig. 6d-f). In addition, to compare
378 CFD results with measured data, the normalized C_2H_6 concentration K is defined as

379 below:

$$380 \quad K = CHU_{ref}/w_{source}dx \quad (14)$$

381 Here the height of building (H), the reference velocity (U_{ref}) and the line source
382 emission strength (w_{source}, dx) are applied.

383 As shown in Fig. 7, the agreements of K between wind tunnel data and CFD
384 results are well confirming the standard $k-\varepsilon$ model has sufficient modeling accuracy in
385 predicting passive pollutant dispersion within 3D urban district models.

386

387 **5 Results and discussion**

388 **5.1 Flow patterns in 3D urban district models**

389 Fig. 8a depicts velocity distribution in the plane of $z=1.5\text{m}$ (the pedestrian level).
390 Obviously, the flow adjustment process can be observed through the entire building
391 clusters, in which wind speed decreases toward downstream from Street 1 to Street 6,
392 and then reaches a comparatively flow equilibrium from Street 7 to Street 18. Fig.
393 8b-c further depict velocity magnitude and 2D streamlines in the plane of $y=30\text{m}$ (the
394 center plane of the target street canyon) and $z=1.5\text{m}$ for Street 3 and Street 10.
395 Moreover, the corresponding 3D streamlines are displayed in Fig. 8d. For both street
396 units, 3D downward helical vortices are produced inside the secondary streets (Fig.
397 8b-d). The lateral flow direction near street ground (i.e. $z=1.5\text{m}$) are from the
398 secondary streets to the main streets and from the downwind building (No.4 and 11)

399 to upwind building (No. 3 and 10) (Fig. 8c). In addition, Street 3 apparently
400 experiences greater wind speed than Street 10.

401 In particular, the following analysis (subsection 5.3) emphasizes more on
402 reactive pollutant dispersion in the fully-developed region (e.g. Street 10) than the
403 flow-adjust region (e.g. Street 3).

404

405 **5.2 Impacts of source locations (S3, S10 and Sm) on reactive pollutant dispersion**

406 This subsection considers the impacts of source locations (i.e. S3, S10 and Sm in
407 Fig. 3) on reactive pollutant dispersion (with chemical reactions, R-type) under the
408 specific pollutant proportion (i.e. $R_{\text{NO}/\text{NO}_2}=100:10$, $[\text{O}_3]=20\text{ppbv}$). Because the present
409 photochemical mechanism contains the interconversion of nitrogen oxides (i.e.
410 $\text{NO}_x=\text{NO}+\text{NO}_2$) and oxidants (i.e. $\text{OX}=\text{NO}_2+\text{O}_3$), The following discussions
411 (subsection 5.2 and 5.3) concentrate more on NO_2 to simplify analysis. Moreover,
412 passive dispersion (without chemical reactions, P-type) are also presented to
413 investigate the sole role of turbulent mixing.

414

415 **5.2.1 NO_2 concentration distribution in 3D urban-like models**

416 Fig. 9a-d exhibit NO_2 concentration between P-type (passive, left) and R-type
417 (reactive, right) cases in $y=30\text{m}$ and $z=1.5\text{m}$ for Street 3 and 10 fixed with span-wise
418 sources (i.e. S3 and S10). Moreover, Fig. 9e compares NO_2 concentration at the

419 pedestrian level between passive and reactive cases with NO_x sources along main
420 streets (i.e. S_m). For passive dispersion with S3 or S10 sources (P-type), due to source
421 emissions and turbulent transports, NO_2 concentration near the upwind building (No.3
422 and 10) is higher than that near the downwind building(Fig. 9a-b). Furthermore, a
423 large amount of NO_2 pollutants accumulate in the intersection of main street and
424 secondary street(Fig. 9c-d). In an overall view, NO_2 concentration in Street 10 is
425 slightly higher than Street 3. For P-type case with S_m sources(Fig. 9e), NO_2
426 concentration first rises toward downstream streets, then reaches an approximate
427 equilibrium from Street 7 to 18. Such findings are similar with the flow adjustment as
428 discussed in subsection 5.1.

429 With chemical reactions, as verified by Fig. 9, NO_2 concentration in R-type cases
430 considerably exceeds that in P-type cases. Oppositely, passive NO concentration is
431 higher than that in R-type cases.

432

433 5.2.2 d_{ps} distribution in Street 3 and 10

434 Fig. 10 shows d_{ps} distribution in $y=30\text{m}$ and $z=1.5\text{m}$ in local target streets with
435 S3 and S10 sources. Here, the distribution of photostationary state defect(d_{ps}) is
436 emphasized below the roof level ($z/H<1$) and toward downstream domains (i.e. $x/H>5$
437 and $x/H>19$ for Street 3 and 10, respectively).

438 As introduced in subsection 2.2, smaller d_{ps} value represents the less departure
439 degree from photochemical equilibrium. In the centre plane of secondary streets

440 ($y=30\text{m}$, Fig. 10a), the local small d_{ps} values emerge near the roof of the upwind
441 building (No.3 and 10) while the large d_{ps} values appear near the roof of the
442 downwind building (No.4 and 11) and the ground level close to NO_x emissions. At the
443 pedestrian level(Fig. 10b), downstream areas of the main streets ($x/H>6$ and $x/H>20$)
444 experience small d_{ps} values while the junction regions of the main street and
445 secondary street near NO_x source locations obtain large d_{ps} values, particularly in
446 Street 3. In summary, d_{ps} value is usually smaller in regions with weaker wind and
447 turbulence, where reactive pollutants have more time to mix and react.

448

449 **5.2.3 Concentration, $\langle P_{iF} \rangle_w$ and $\langle P_{iF} \rangle_B$ on building wall surfaces**

450 Based on spatial mean concentration at the entire building surfaces, we calculate
451 wall intake fraction($\langle P_{iF} \rangle_w$) and building intake fraction($\langle P_{iF} \rangle_B$) to analyze
452 overall vehicular pollutant(NO_x) exposure in near-road buildings. Especially, 1ppm
453 represents 1 mg inhaled averagely by each person living in the near-road building if
454 1kg pollutants emitted out.

455 Fig. 11a-b first compare NO_2 concentration on the leeward and windward walls
456 between P-type and R-type cases in target street units with S3 and S10 sources. No
457 matter with or without chemical reactions, NO_2 concentrations on the leeward wall
458 are always higher than those on the windward wall. Once $\text{NO}_x\text{-O}_3$ photochemical
459 reactions are conducted, an increase of NO_2 concentration emerges on the upwind and
460 downwind walls. Such results are similar with the aforementioned discussion in

461 subsection 5.2.1.

462 Then, Table 4 lists $\langle P_{iF} \rangle_w$ of NO and NO₂ at leeward and windward walls
463 adjoining target street in cases with S3 and S10 (positions as described in Fig. 3a-b).
464 Obviously, in both P-type and R-type cases, $\langle P_{iF} \rangle_w$ of NO_x at leeward wall (Table 4,
465 row 1-4, column 2-3) are greater than those at windward wall (Table 4, row 1-4,
466 column 4-5). Regarding P-type cases as the references, $\langle P_{iF} \rangle_w$ of NO₂ in R-type
467 cases rise nearly 90%-160%, i.e. 0.660 to 1.372ppm and 0.853 to 1.643ppm at
468 leeward wall for S3 and S10 (Table 4, column 2), 0.180 to 0.473ppm and 0.230 to
469 0.610ppm at windward wall for S3 and S10 (Table 4, column 4); while $\langle P_{iF} \rangle_w$ of
470 NO reduces about 9%-16%, i.e. 0.660 to 0.588ppm and 0.853 to 0.775ppm at leeward
471 wall for S3 and S10 (Table 4, column 3), 0.180 to 0.151ppm and 0.230 to 0.193ppm at
472 windward wall for S3 and S10 (Table 4, column 5). Because NO_x emission ratio
473 released from vehicles is $R_{NO/NO_2}=10$, the present photochemical processes lead to a
474 significant increase in NO₂ exposure and a slight decrease in NO exposure.

475 Furthermore, NO₂ concentrations on the entire building wall surfaces with Sm
476 sources are presented in Fig. 11c. Both P-type and R-type cases experience the NO₂
477 concentration adjustment processes toward downstream buildings. To quantify NO_x
478 exposure adjustments in P-type and R-type cases with various sources (S3, S10 and
479 Sm), the horizontal profiles of building intake fraction $\langle P_{iF} \rangle_B$ of NO and NO₂ are
480 shown in Fig. 12. It is found that $\langle P_{iF} \rangle_B$ with S3 or S10 sources descends
481 exponentially toward downstream buildings (Fig. 12a-b), instead, $\langle P_{iF} \rangle_B$ with Sm
482 sources first ascends quickly from building No.1 to 8, then reaches an approximate

483 equilibrium(Fig. 12c). Besides, R-type cases experience larger NO₂ exposure than
 484 P-type cases, i.e. 0.420-0.108ppm against 0.135-0.020ppm for S3(Fig. 12a),
 485 0.605-0.160ppm against 0.207-0.030ppm for S10(Fig. 12b), 0.005-0.090ppm against
 486 0.002-0.049ppm for Sm(Fig. 12c). Oppositely, NO exposure in P-type cases are
 487 greater than R-type cases, i.e. 0.135-0.020ppm than 0.106-0.010ppm for S3(Fig. 12a),
 488 0.207-0.030ppm than 0.168-0.017ppm for S10(Fig. 12b) and 0.002-0.049ppm than
 489 0.002-0.045ppm for Sm(Fig. 12c).

490 In addition, the decay function expressed in $\langle P_{iF_n} \rangle_B = a \times \langle P_{iF_t} \rangle_B \times e^{(t-n)/b}$ is
 491 employed to further quantify the $\langle P_{iF} \rangle_B$ decay processes from target building unit (t
 492 =4 or 11) toward downstream building ($n=t$ to 21) in S3 and S10 cases. Note that,
 493 smaller decay factor b means relatively sharper descending processes of $\langle P_{iF} \rangle_B$
 494 curves. Table 5 summarizes the $\langle P_{iF_t} \rangle_B$ of building “No. t ” and the exponential
 495 decay factors b in S3 and S10 cases. Obviously, compared with those in P-type case
 496 ($b=7.46$ and 3.96 in Table 5, row 1 and 3), R-type cases with S3 and S10 obtain larger
 497 b for NO₂ (10.80 and 6.84 in Table 5, row 2 and 4) and smaller b for NO (5.92 and
 498 2.91). Moreover, $\langle P_{iF} \rangle_B$ curves in S10 cases(Table 5, row 3-4) decline more
 499 sharply toward downstream regions than S3 cases(Table 5, row 1-2). Particularly, as
 500 shown in Fig. 12c, $\langle P_{iF_t} \rangle_B$ in Sm cases are calculated by the mean $\langle P_{iF} \rangle_B$ from
 501 building No.9 to 21, i.e. 0.048ppm in P-type case, 0.044ppm and 0.088ppm in R-type
 502 case for NO and NO₂.

503 In summary, with $R_{NO/NO_2}=100:10$ and $[O_3]=20$ ppbv, the present NO_x-O₃
 504 titration interactions result in the production of NO₂ and depletion of O₃ and NO. By

505 focusing on $\langle P_{iF} \rangle_B$ values, NO₂ exposure in R-type cases are greater than P-type
506 cases about 3.1 times for S3, 2.9 times for S10 and 1.8 times for Sm while NO
507 exposure in R-type cases are nearly 21%, 19% and 8% smaller than P-type cases for
508 S3, S10 and Sm, respectively.

509

510 **5.3 Impacts of reactant proportions (NO:NO₂:O₃) on reactive pollutant dispersion**

511 In this subsection, based on S10 sources, we discuss the effects of different
512 reactant proportions (NO:NO₂:O₃) on the interaction of turbulent mixing and
513 photochemical processes in urban districts. Additionally, P_{iF} and $\langle P_{iF} \rangle$ are
514 independent on source emission strength in passive pollutant dispersion, therefore
515 Case P[S10,100:10] is treated as the reference case to compare with the cases with
516 other reactant proportions.

517

518 **5.3.1 Impacts of O₃ background concentration ([O₃])**

519 With the same emission ratio of NO to NO₂ (i.e. $R_{NO/NO_2}=100:10$), the impacts of
520 four O₃ background concentrations (i.e. [O₃] =1, 20, 40 and 100ppbv) on reactive
521 pollutant dispersion are examined.

522 It is apparent that the formation of NO by photolyzing NO₂ is slightly dominant in
523 photochemistry when [O₃] is 1ppbv. In contrast to the reference values (0.853 and
524 0.230ppm in [Table 4, row 3](#)), $\langle P_{iF} \rangle_w$ of NO₂ at the leeward and windward walls

525 slightly decrease(i.e. 0.816 and 0.211ppm in Table 4, row 5) while $\langle P_{iF} \rangle_w$ of NO
526 increase a little(i.e. 0.858 and 0.233ppm). Besides, such phenomenon is distinctly
527 observed in $\langle P_{iF} \rangle_B$ curves(Fig. 13) between the reference case and Case R
528 [S10,100:10,1], i.e. 0.207-0.030ppm against 0.187-0.020ppm for NO₂(Fig. 13a) and
529 0.207-0.030ppm against 0.209-0.031ppm for NO(Fig. 13b).

530 However, if [O₃] rises from 20ppbv to 40 and 100ppbv, more NO is oxidized by
531 O₃ to generate NO₂. Based on the reference values (0.853 and 0.230ppm in Table 4,
532 row 3), $\langle P_{iF} \rangle_w$ of NO₂ become about 1.9-5.2 times greater (1.643, 2.454 and
533 4.442ppm in Table 4, column 2) at leeward wall and 2.6-6.7 times larger (0.610, 0.944
534 and 1.534ppm in Table 4, column 4) at windward wall; while $\langle P_{iF} \rangle_w$ of NO
535 reduces approximately 9%-42% (0.775, 0.694 and 0.495ppm in Table 4, column 3) at
536 leeward wall and 16%-57% (0.193, 0.159 and 0.100ppm in Table 4, column 5) at
537 windward wall. Furthermore, Fig. 13 and Table 5 display the corresponding $\langle P_{iF} \rangle_B$
538 curves, $\langle P_{iF} \rangle_B$ values and decay factors b under different [O₃]. As depicted in Fig.
539 13, compared with Case R[S10,100:10,20], Case R[S10,100:10,100] and R
540 [S10,100:10,40] obviously attain much larger $\langle P_{iF} \rangle_B$ of NO₂ (1.573-0.275 and
541 0.952-0.218ppm in Fig. 13a) and smaller $\langle P_{iF} \rangle_B$ of NO (0.071-0.005 and
542 0.133-0.011ppm in Fig. 13b). Additionally, both decay factors of NO₂ and NO are
543 smaller ($b=6.02$ and 2.33 for [O₃]=40ppbv, $b=4.74$ and 1.87 for [O₃]=100ppbv in
544 Table 5, row 6-7) than those in Case R[S10,100:10,20] ($b=6.84$ and 2.91 in Table 5,
545 row 4), which implies higher [O₃] induces the quicker decay of $\langle P_{iF} \rangle_B$ curves for
546 NO_x with S10 sources toward downstream building units. By concentrating on

547 $\langle P_{iF} \rangle_B$, NO₂ exposure in R-type cases surpass the reference case nearly 2.9, 4.6 and
548 7.6 times for [O₃]=20, 40 and 100ppbv, respectively (Table 5, column 2).
549 Correspondingly, $\langle P_{iF} \rangle_B$ of NO in these [O₃] cases are about 19%, 36% and 66%
550 smaller than the reference values, respectively (Table 5, column 4). It clearly indicates
551 that increasing [O₃] would aggravate NO₂ exposure within urban clusters but is
552 conducive to the mitigation of NO exposure.

553

554 **5.3.2 Effects of emission ratio of NO to NO₂ (R_{NO/NO_2})**

555 The effects of source emission ratios (R_{NO/NO_2} =100:10, 50:10 and 100:20) on
556 reactive pollutant dispersion are investigated with the same [O₃] of 20ppbv.

557 It is shown that decreasing NO or increasing NO₂ emissions based on the
558 reference case can mildly change the fraction of NO converting into NO₂. For
559 example, reducing R_{NO/NO_2} from 100:10 to 50:10 and 100:20, $\langle P_{iF} \rangle_w$ of NO
560 varies from 0.775 to 0.744 and 0.784ppm at the leeward wall (Table 4, column 3), and
561 from 0.193 to 0.186 and 0.197ppm at windward wall (Table 4, column 5). Furthermore,
562 $\langle P_{iF} \rangle_w$ of NO₂ drops from 1.643 to 1.406 and 1.205ppm at the leeward wall (Table
563 4, column 1), and from 0.610 to 0.453 and 0.400ppm at windward wall (Table 4,
564 column 4). In addition, Fig. 14 presents $\langle P_{iF} \rangle_B$ curves of NO₂ and NO in P-type and
565 R-type cases with three NO-NO₂ emission ratios. Obviously, photochemical reactions
566 in these R-type cases are still dominated by the depletion of O₃ with NO to produce
567 NO₂.

568 As displayed in Fig. 14a and Table 5, Case R[S10,50:10,20] and R
569 [S10,100:20,20] obtain smaller $\langle P_{iF} \rangle_B$ and decay factor of NO₂ (i.e. 0.446-0.091
570 ppm, $b=5.41$ and 0.384-0.088ppm, $b=6.00$) than those in Case R[S10,100:10,20] (i.e.
571 0.605-0.160ppm, $b=6.84$). In contrast to Case R[S10,100:10,20], $\langle P_{iF} \rangle_B$ of NO in
572 Case R[S10,50:10,20] and R[S10,100:20,20] reduce nearly 26% and 37%. However,
573 $\langle P_{iF} \rangle_B$ curves and decay factors b of NO are quite close between three R-type cases,
574 i.e. 0.172-0.018, 0.168-0.018 and 0.160-0.017ppm; $b=3.05$, 2.91 and 3.11(Fig. 14b
575 and Table 5). It is confirmed that the decrement of R_{NO/NO_2} (from 10 to 5) can partly
576 offset NO₂ exposure but have much less impacts on NO exposure.

577 Overall, the NO_x-O₃ photochemical processes dependent on the initial proportion
578 of reactive pollutants are toward satisfying the photostationary state relationship
579 (i.e. $k_f[NO][O_3]=J_{NO_2}[NO_2]$).

580

581 **5.4 Limitations and future work**

582 Since the 3D urban district models, photochemical reactions and meteorological
583 conditions adopted in this study are fairly simplified, the present exposure results may
584 change if more realistic factors are taken into account, such as more realistic urban
585 configurations(e.g. with variations of building height and street width), more
586 complicated chemical mechanisms(e.g. VOCs-NO_x-O₃) and more realistic
587 atmospheric conditions etc. It is worth mentioning that the chemical processes
588 dependent on reaction rates are highly associated with the reactive pollutant

589 concentration and ambient air temperature. Moreover, realistic atmospheric conditions
590 include the unsteady temporal and spatial variations of wind speed and direction as
591 well as various atmospheric stabilities and solar radiation conditions. Thus, further
592 unsteady CFD simulations will be performed to examine the integrated impacts of
593 urban turbulence and solar radiation on reactive pollutant dispersion in 3D urban
594 districts.

595

596 **6 Conclusions**

597 Urban residents in near-road buildings commonly suffer from high exposure risk
598 of vehicular pollutants in which NO_x (NO and NO_2) act as primary reactive pollutants.
599 With satisfactory full-scale CFD validation of flow and pollutant dispersion by
600 experimental data, this study first focuses on the impact of turbulent transport
601 combined with NO_x - O_3 photochemical reactions on reactive pollutant dispersion in
602 neighborhood-scale (21-row, $\sim 1\text{km}$) three-dimensional (3D) medium-dense urban
603 clusters ($H/W=1$, $\lambda_p=\lambda_f=0.25$). Ground-level emission sources of NO and NO_2 are
604 considered in the presence of background O_3 . The approaching wind is parallel to the
605 main streets and perpendicular to the secondary streets. As a start, the influences of
606 various source locations and reactant proportions ($\text{NO}:\text{NO}_2:\text{O}_3$) on pollutant dispersion
607 are investigated under neutral meteorological condition. Personal intake fraction P_{iF} ,
608 its spatially-averaged values for a building wall ($\langle P_{iF} \rangle_w$) and the entire building
609 surfaces (i.e. building intake fraction $\langle P_{iF} \rangle_B$) are adopted to quantify pollutant

610 exposure with and without NO-NO₂-O₃ reactions(i.e. reactive and passive).

611 Some meaningful findings are summarized as below:

612 1) There are flow adjustment processes coupling turbulent mixing and chemical
613 reactions through urban building clusters(Street 1 to Street 20 toward
614 downstream).With span-wise sources, the secondary street of Street 10 located in the
615 fully-developed region(i.e. S10 case) experiences weaker wind and subsequently
616 greater $\langle P_{iF} \rangle_B$ than the secondary Street 3 located in the upstream flow-adjustment
617 region (i.e. S3 case). Consequently, in contrast to S3 case, photostationary state defect
618 (d_{ps}) is smaller in S10 case since reactive pollutants have more time to mix and react
619 in Street 10.

620 2) With source emission ratios of NO to NO₂ of 10($R_{NO/NO_2}=100:10$) and
621 background O₃ concentration of 20ppbv($[O_3]=20ppbv$), NO-NO₂-O₃ photochemistry
622 leads to production of NO₂ and depletion of O₃ and NO, inducing a significant
623 increase in NO₂ exposure and a slight decrease in NO exposure when compared to
624 corresponding passive dispersion which only considers the sole role of turbulent
625 transport. With span-wise pollutant sources, 3D downward helical flows transport
626 more NO_x to the leeward side, inducing much greater leeward-side $\langle P_{iF} \rangle_w$ than the
627 windward-side. Moreover, by defining exponential decay function expressed
628 in $\langle P_{iF}_n \rangle_B = a \times \langle P_{iF}_t \rangle_B \times e^{(t-n)/b}$, it is found that $\langle P_{iF} \rangle_B$ descends exponentially
629 from target building ($\langle P_{iF} \rangle_B=0.135ppm$ or $0.207ppm$, $t=4$ or 11 for S3 or S10) to
630 downstream buildings ($n=t$ to 21). Especially, $\langle P_{iF} \rangle_B$ curves decline more sharply

631 from Street 10 toward downstream than that from Street 3. However, if stream-wise
632 sources fixed along the main streets, $\langle P_{iF} \rangle_B$ first ascends quickly from building
633 No.1 to 8, then reaches approximate equilibrium values of $\langle P_{iF} \rangle_B =$
634 0.046-0.049ppm.

635 3) Furthermore, the O_3 background concentration ($[O_3]$) and source emission
636 ratios of NO to NO_2 (R_{NO/NO_2}) are confirmed as key factors on NO_x-O_3 reactive
637 dispersion. The formation of NO by photolyzing NO_2 is slightly dominant in
638 photochemistry when $[O_3]$ is 1ppbv. However, if $[O_3]$ rises from 20ppbv to 40 and
639 100ppbv, more NO is oxidized by O_3 to generate NO_2 , which would aggravate NO_2
640 exposure within urban clusters but is conducive to the mitigation of NO exposure.
641 Under $[O_3]$ of 20ppbv, results show that the decrement of R_{NO/NO_2} from 10 to 5 can
642 partly offset NO_2 exposure but have much less impacts on NO exposure.

643 Although further investigations are still required to provide practical guidelines,
644 this paper is one of the first attempts to quantify how reactive pollutant source
645 locations and reactant proportions influence reactive pollutant exposure in 3D urban
646 districts, which can present meaningful references for urban planning. The effective
647 methodologies are proposed for reactive pollutant exposure assessment in more
648 complicated urban districts with various meteorological conditions and chemical
649 mechanisms.

650

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936

937 **Figure list:**

938 Fig.1 (a) The population census data of Hong Kong [4]; (b) Time activity patterns for
939 each subgroup in four microenvironments [70].

940 Fig.2 (a) Idealized 3D urban district models ($H/W=1$, $\lambda_p=\lambda_f=0.25$); (b)
941 Computational domains and boundary conditions for test cases; (c) The overhead
942 and lateral views of mesh distribution.

943 Fig.3 Model setups of vehicular emission sources: (a) S3, (b) S10 and (c) Sm.

944 Fig.4 (a-b) The side and top views of measured points and model configurations in
945 wind tunnel experiment [87]; (c) Computational domains and boundary
946 conditions in CFD validation study; (d) The mesh arrangements in grid
947 sensitivity test.

948 Fig.5 Results of grid independence test: (a) $\bar{u}(z)$, (b) $\bar{w}(z)$ at Point V1; Comparison of
949 vertical profiles between wind tunnel data and CFD results: (c-e) $\bar{u}(z)$ at Points
950 V1, V4 and V6, respectively; (f-g) $\bar{w}(z)$ and $k(z)$ at Point V1.

951 Fig.6 (a-b) The lateral and overhead views of experiment settings in concentration
952 measurement [92]; (c) CFD setups in passive dispersion validation; (d-f) Vertical
953 profiles of stream-wise velocity, turbulence kinetic energy and turbulent
954 dissipation rate in domain inlet.

955 Fig.7 Comparison of K between wind tunnel data and CFD results applying standard
956 $k-\varepsilon$ model: (a) on the leeward and windward walls; (b) on the central line of roof
957 surface.

958 Fig.8 (a) Velocity distribution in the pedestrian level of $z=1.5\text{m}$; (b-c) Velocity
959 magnitude and 2D streamlines for Street 3 (left) and 10 (right) in the plane of
960 $y=30\text{m}$ and $z=1.5\text{m}$, respectively; (d) 3D streamlines in Street 3 and 10.

961 Fig.9 NO_2 concentration distribution between P-type (left or above) and R-type (right
962 or below) cases: (a-b) in the plane of $y=30\text{m}$ with S3 and S10, respectively; (c-d)
963 in the pedestrian level of $z=1.5\text{m}$ with S3 and S10, respectively; (e) in the
964 pedestrian level of $z=1.5\text{m}$ with Sm.

965 Fig.10 (a-b) Photostationary state defect between S3 (left) and S10 (right) cases in
966 $y=30\text{m}$ and $z=1.5\text{m}$, respectively.

967 Fig. 11 NO_2 concentration between P-type (left or above) and R-type (right or below)
968 cases: (a-b) on the leeward and windward walls with S3 and S10, respectively; (c)
969 on the entire building walls with Sm.

970 Fig.12 $\langle P_{iF} \rangle_B$ curves of NO and NO_2 in P-type and R-type cases with various
971 source locations: (a) S3, (b) S10 and (c) Sm, respectively.

972 Fig. 13 $\langle P_{iF} \rangle_B$ curves of (a) NO₂ and (b) NO in P-type and R-type cases under four
973 O₃ background concentrations ([O₃]=1,20,40,100ppbv).

974 Fig. 14 $\langle P_{iF} \rangle_B$ curves of (a) NO₂ and (b) NO in P-type and R-type cases with three
975 kinds of NO-NO₂ emission ratios.

976

977 **Table list:**

978 Table 1 Time patterns and breathing rates in each subgroup for indoors at home ($j=1$)

979 Table 2 Summary of all test cases

980 Table 3 Comparison between wind tunnel experiment and CFD simulation

981 Table 4 $\langle P_{iF} \rangle_W$ (unit: ppm) at leeward and windward walls adjoining the target
982 street in S3 and S10 cases

983 Table 5 $\langle P_{iF} \rangle_B$ values and the exponential factor b in S3 and S10 cases

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